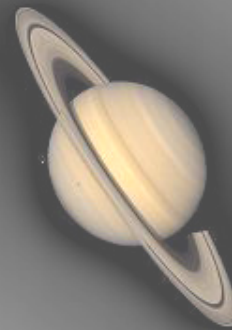
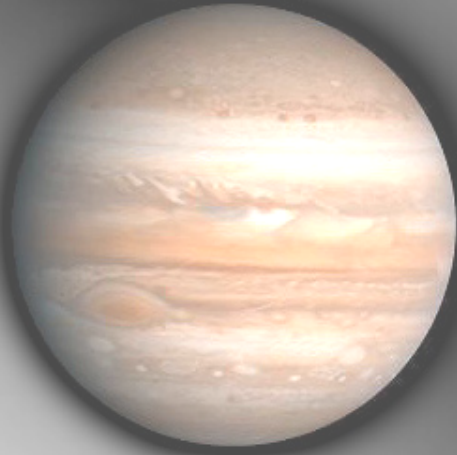


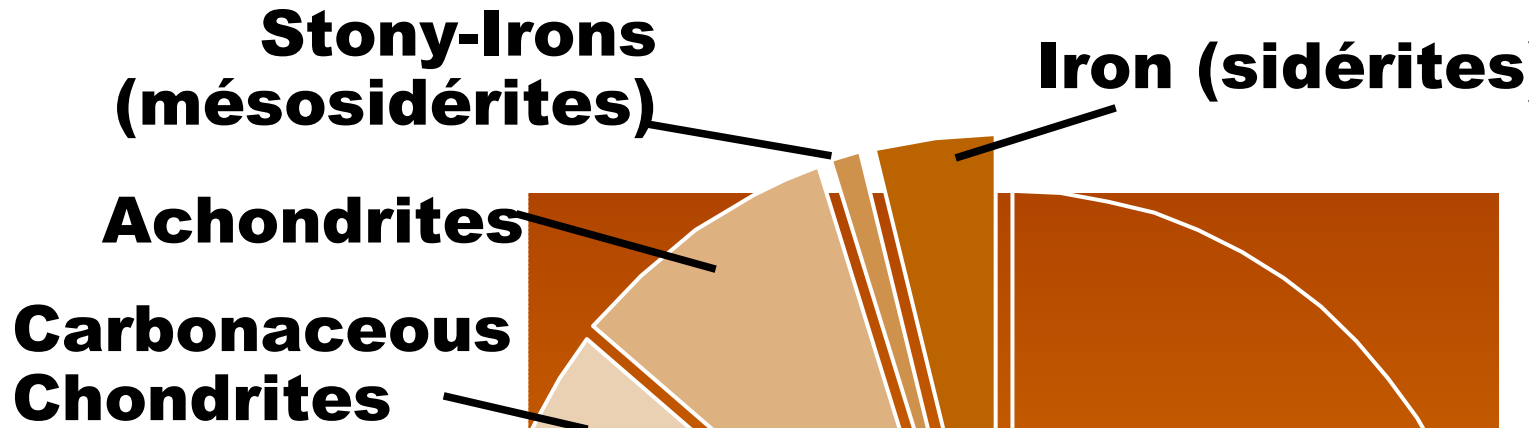
METEORITES, ISOTOPIC DATATION



Aurélien CRIDA

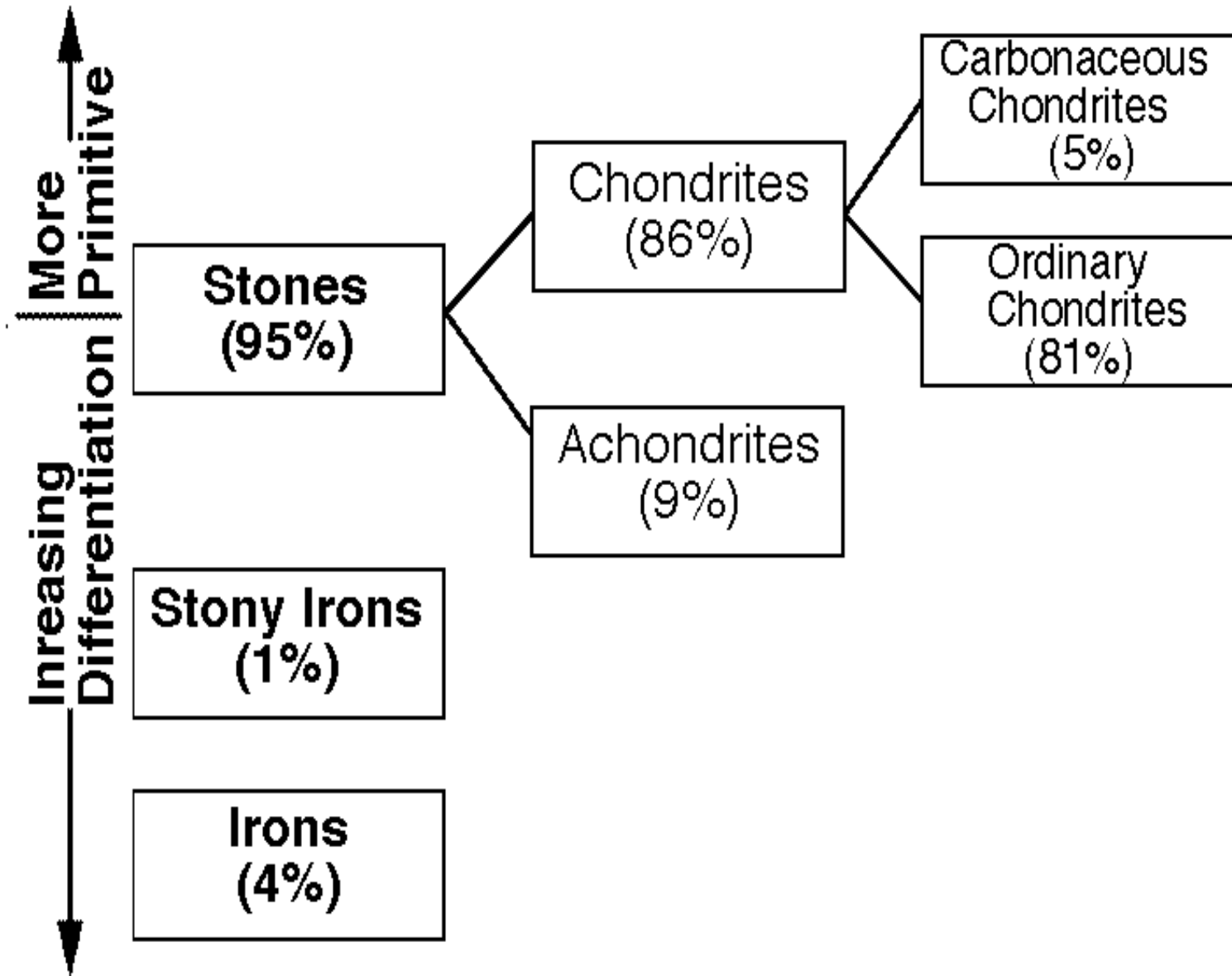
MÉTÉORITES

Falls :



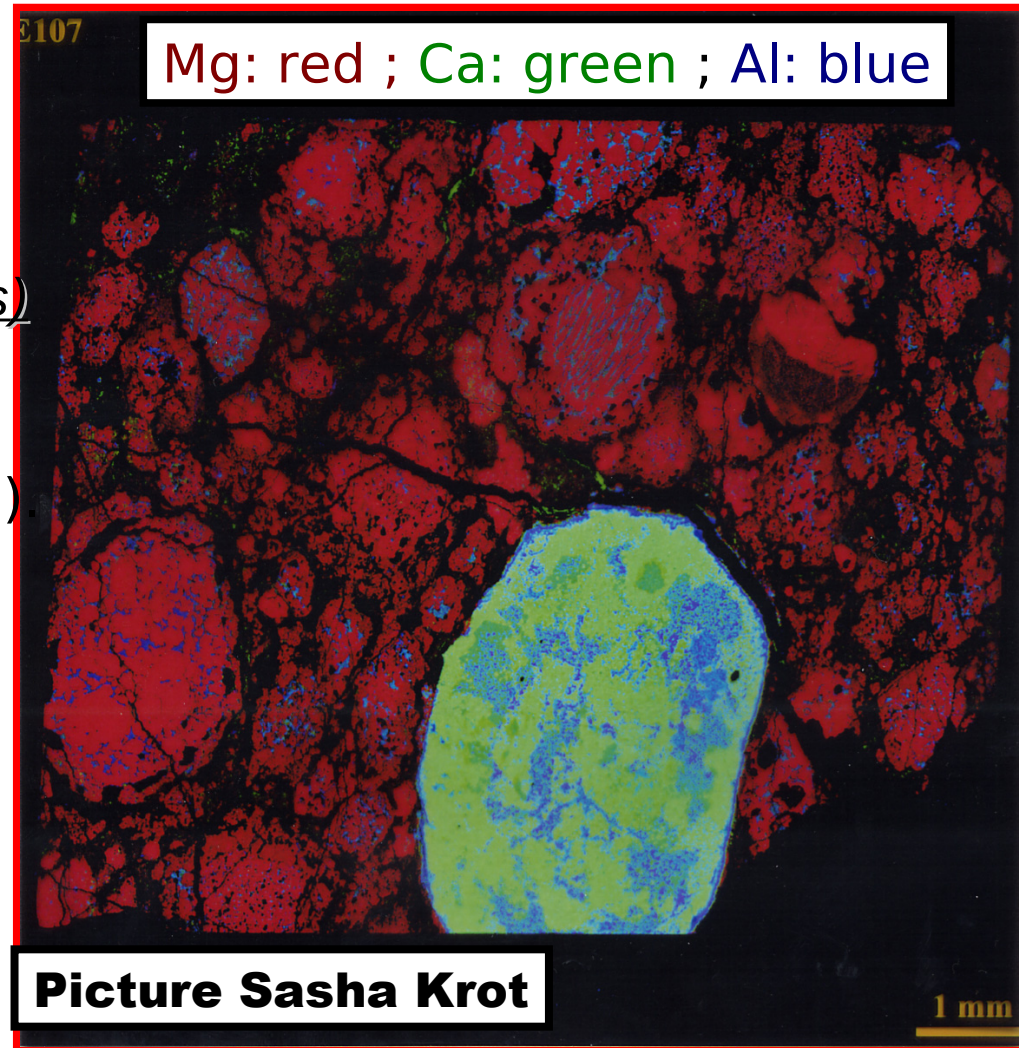
Ordinary Chondrites

MÉTÉORITES : CLASSIFICATION



COMPOSITION of CHONDRITES

- ☆ Chondrules (fr: chondres)
Silicate/metal beads.
Varying sizes (10 μm - few mm).
- ☆ Refractory inclusions = CAI
(Calcium-Aluminum rich Inclusions)
Contain Ca- and Al-rich minerals.
Varying sizes (10 μm - few cm)
- ☆ Matrix (fr: matrice)
Silicates, oxides, amorphous.
Fine-grained (< μm).

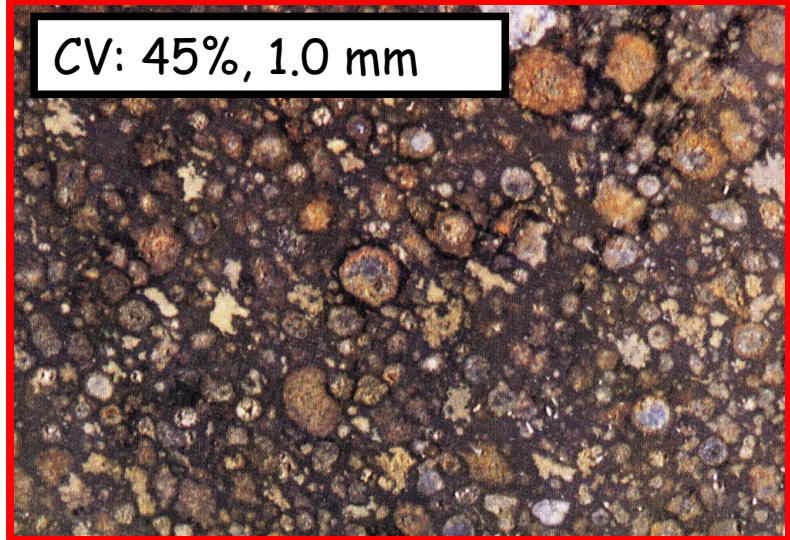


CHONDRULES : Proportion and Size

CM: 20%, 0.3 mm



CV: 45%, 1.0 mm



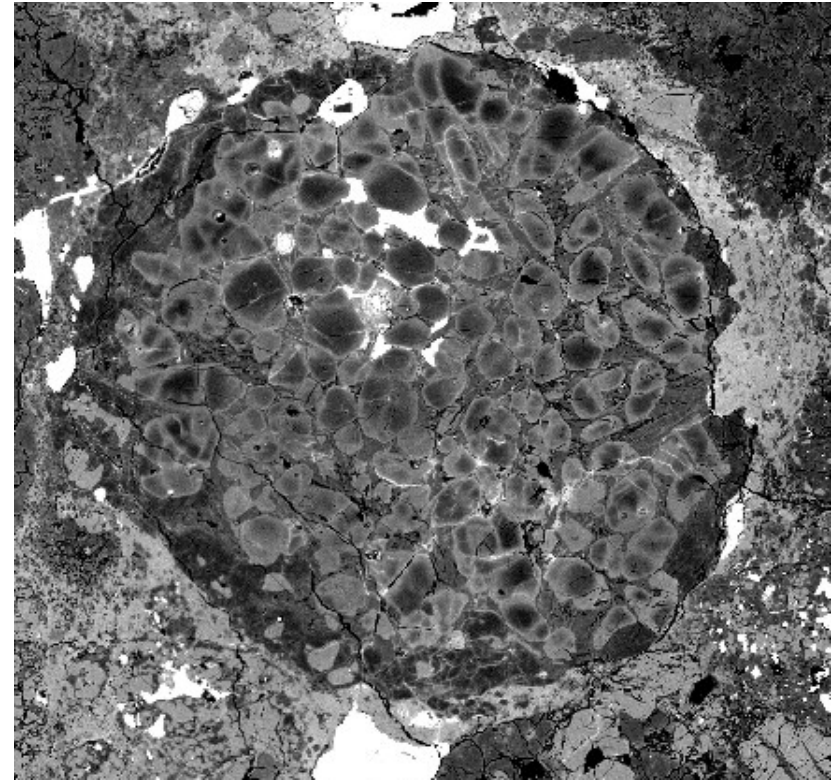
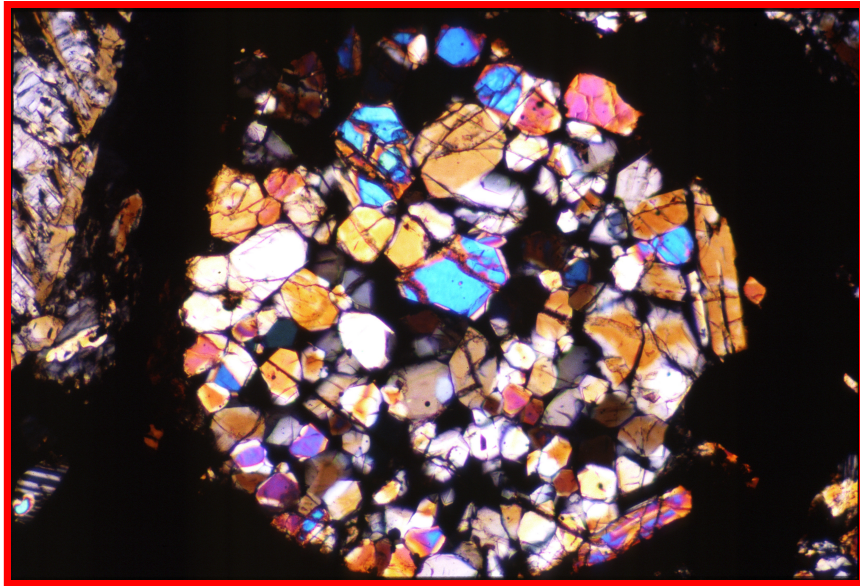
L: 80%, 0.7 mm



CI: 0 %



Ex: PORPHYRYTIC CHONDRULE



- ☆ Olivine $[(Mg,Fe)_2SiO_4]$ and pyroxene $[(Mg,Fe)SiO_3]$ *phenocrysts* set in a glassy matrix

MATRIX of the CHONDRITES

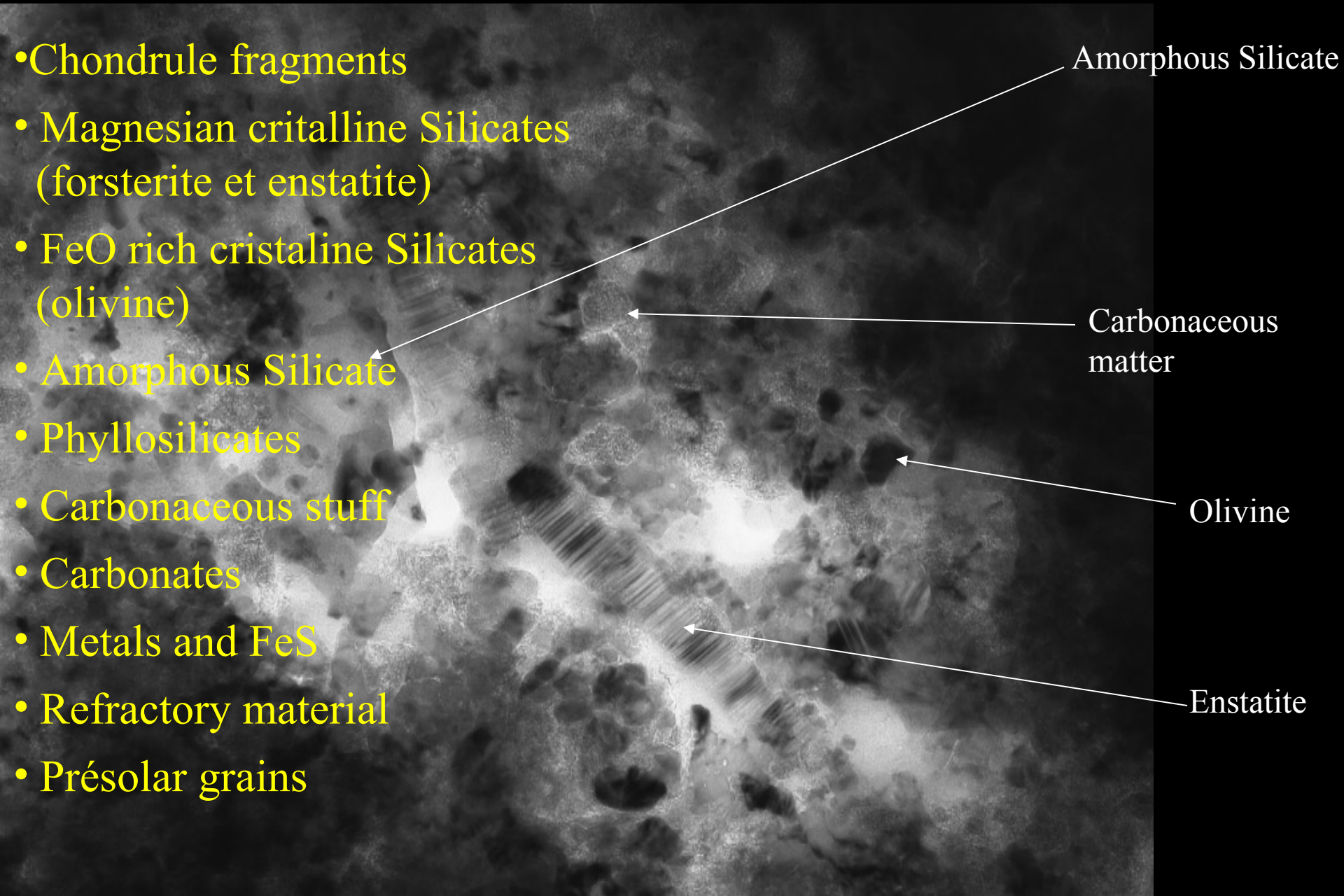
- Chondrule fragments
- Magnesian crystalline Silicates (forsterite et enstatite)
- FeO rich crystalline Silicates (olivine)
- Amorphous Silicate
- Phyllosilicates
- Carbonaceous stuff
- Carbonates
- Metals and FeS
- Refractory material
- Présolar grains

Amorphous Silicate

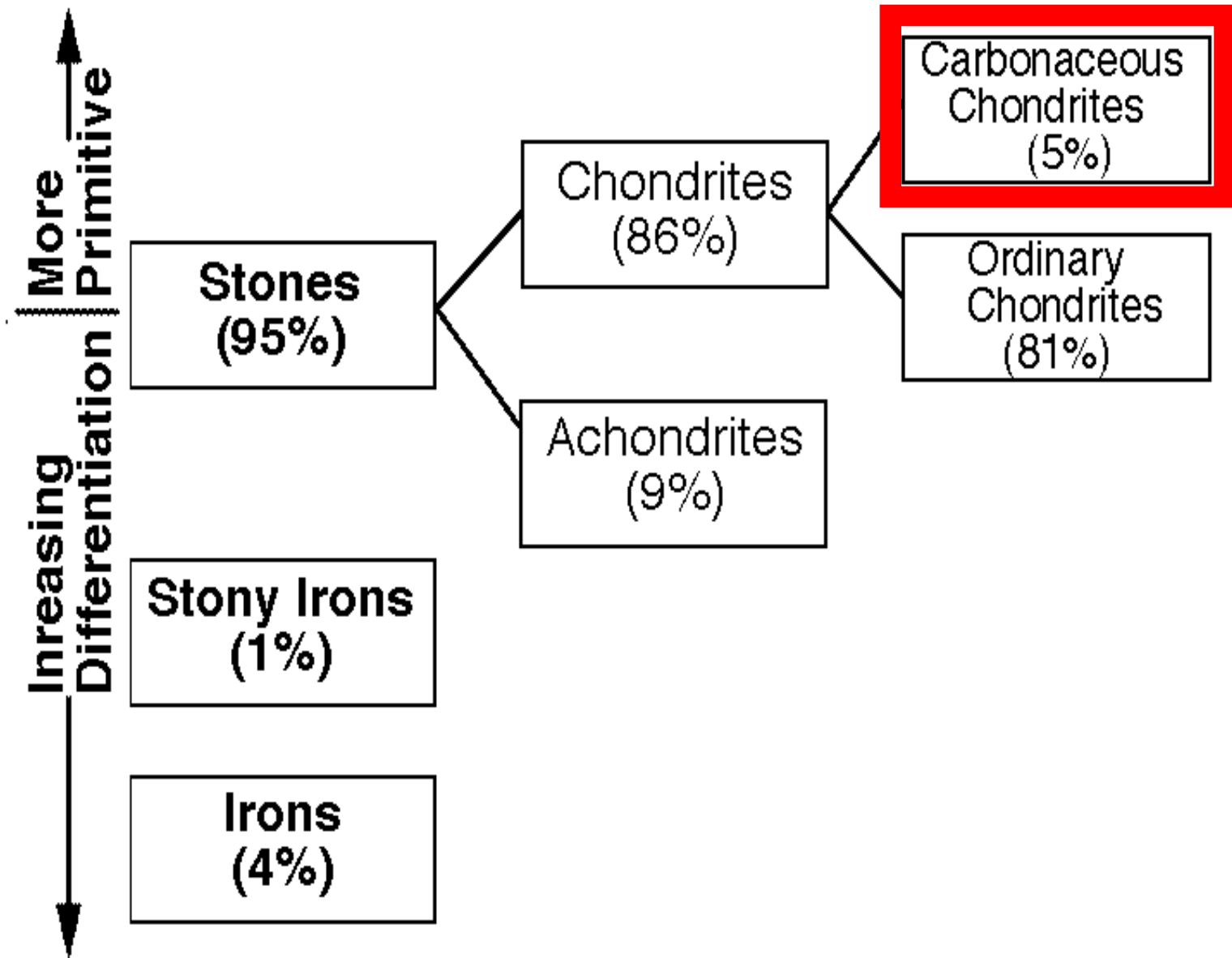
Carbonaceous matter

Olivine

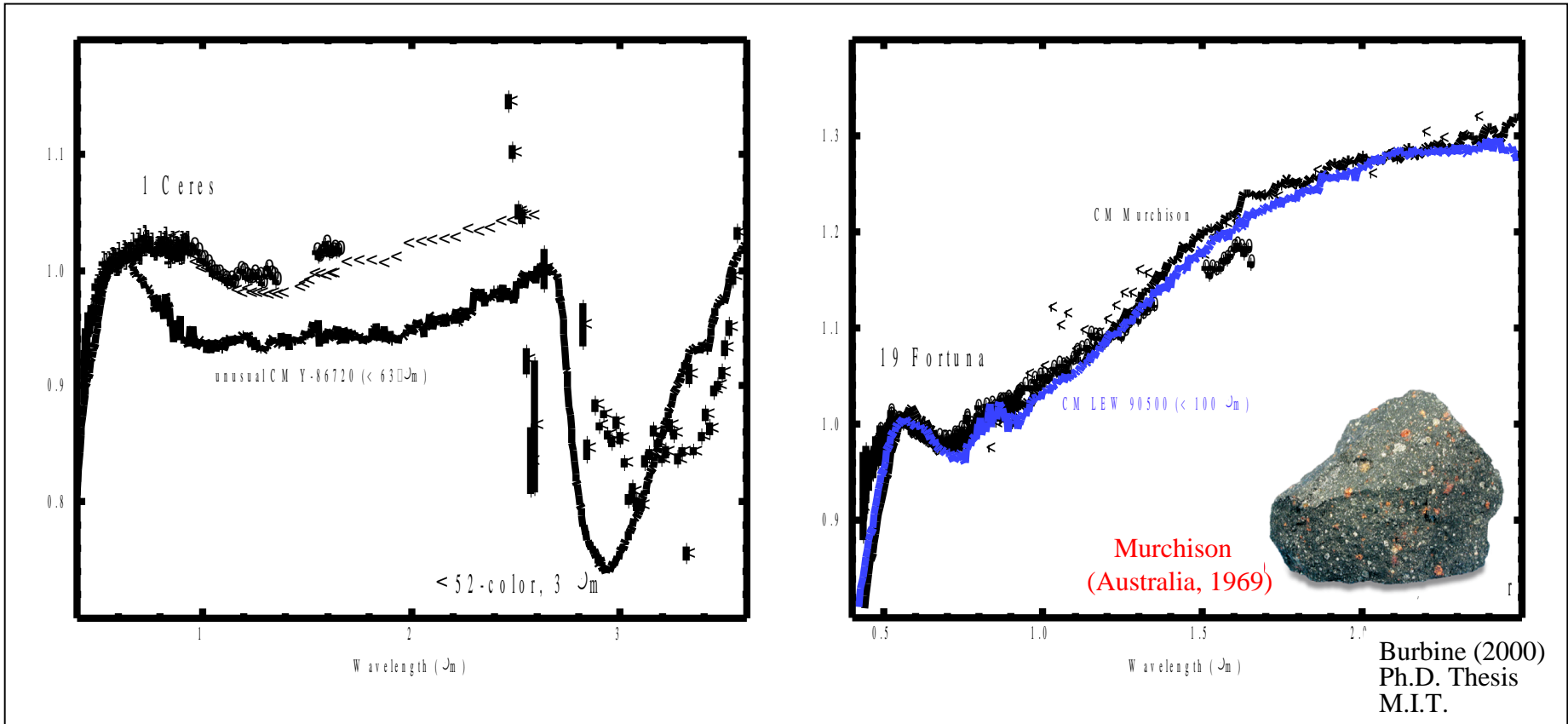
Enstatite



CONNECTIONS ASTÉROÏDES-MÉTÉORITES

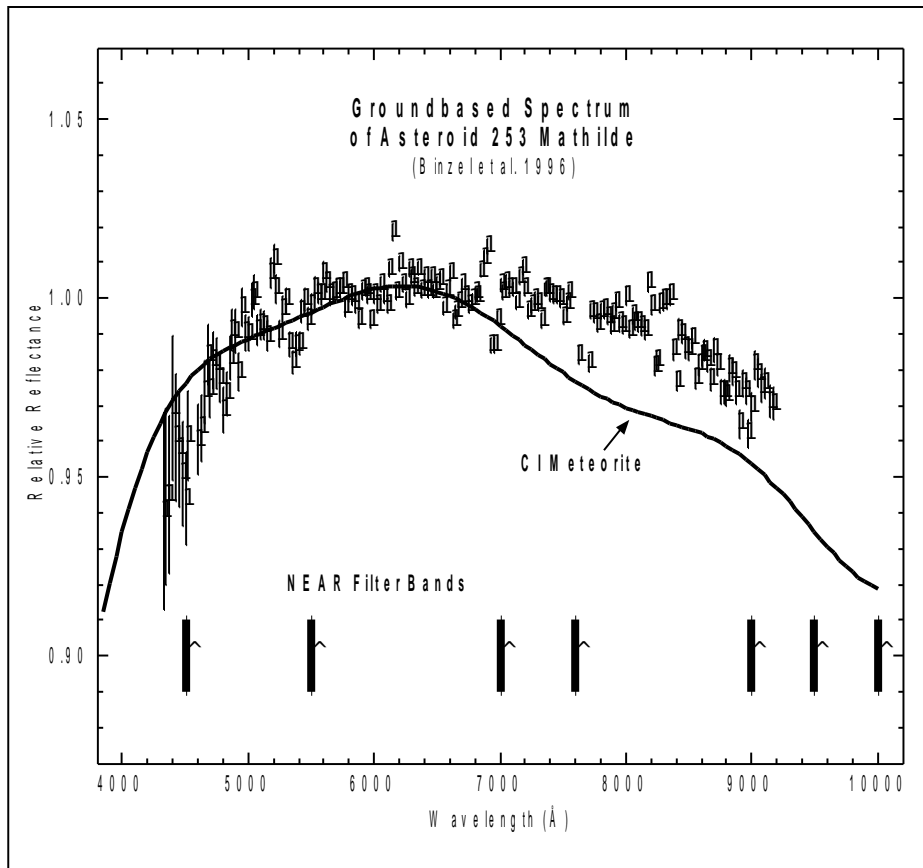


CONNECTIONS ASTÉROÏDES-MÉTÉORITES

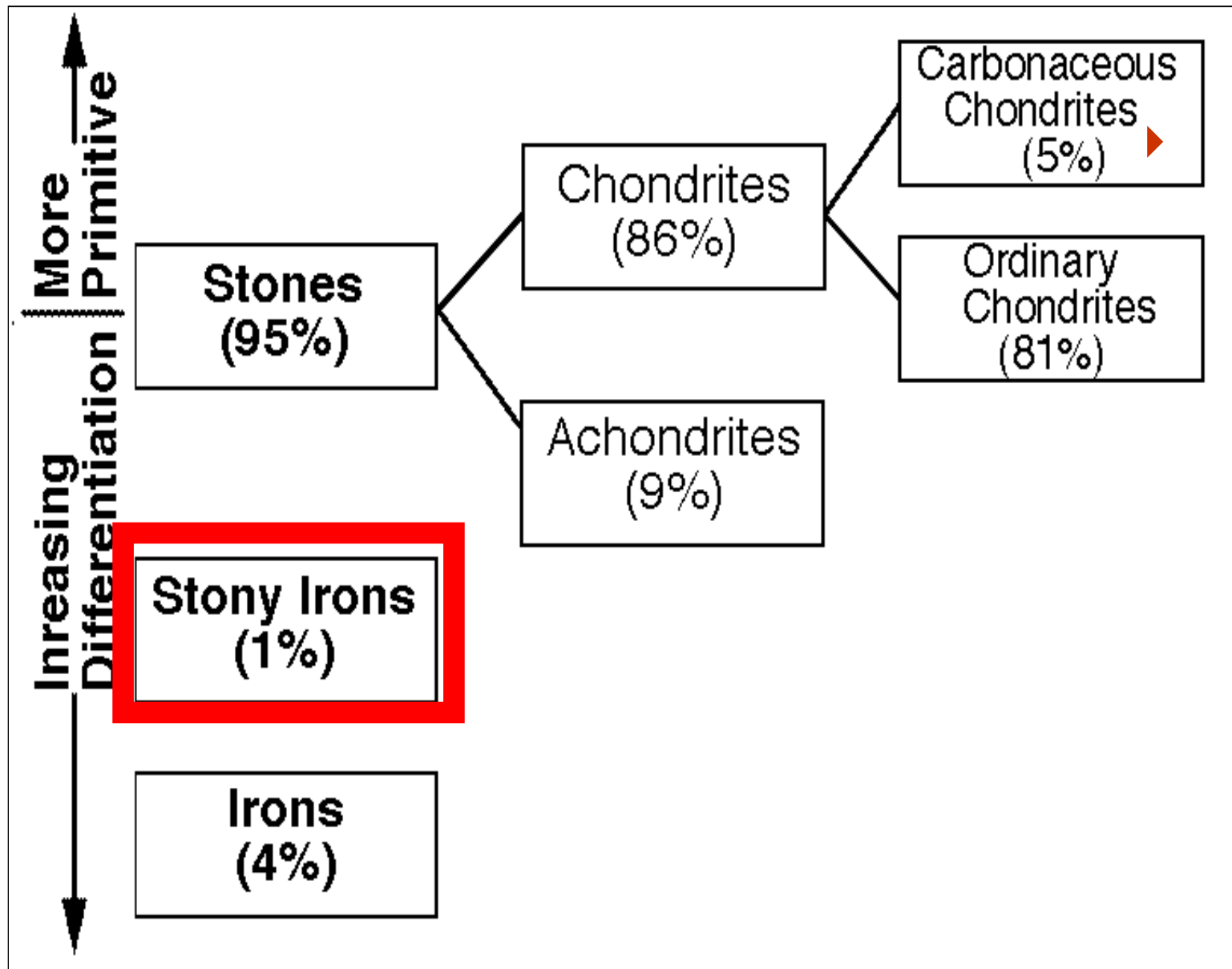


Good agreement between asteroids of spectral type C and Carbonaceous Chondrites.

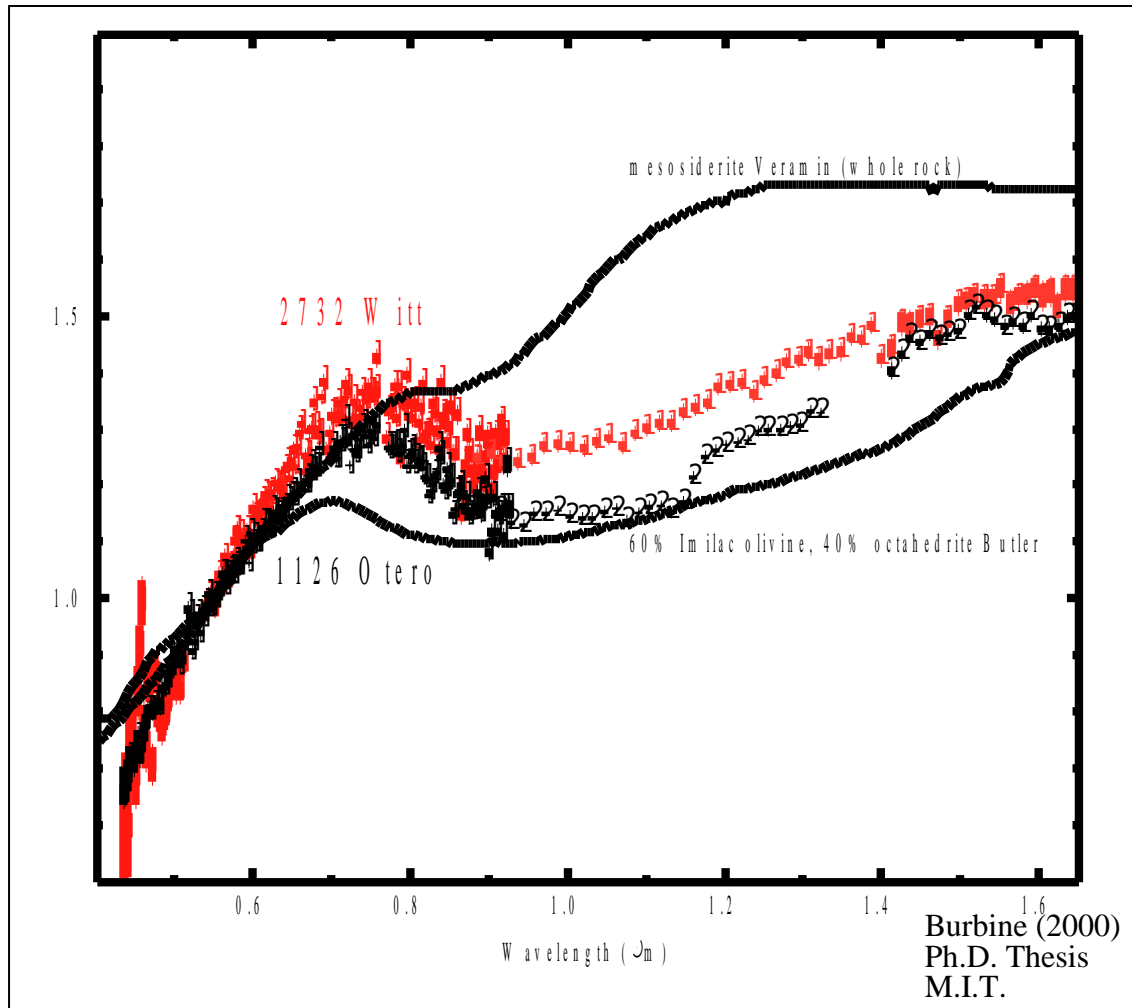
CONNECTIONS ASTÉROÏDES-MÉTÉORITES



CONNECTIONS ASTÉROÏDES-MÉTÉORITES



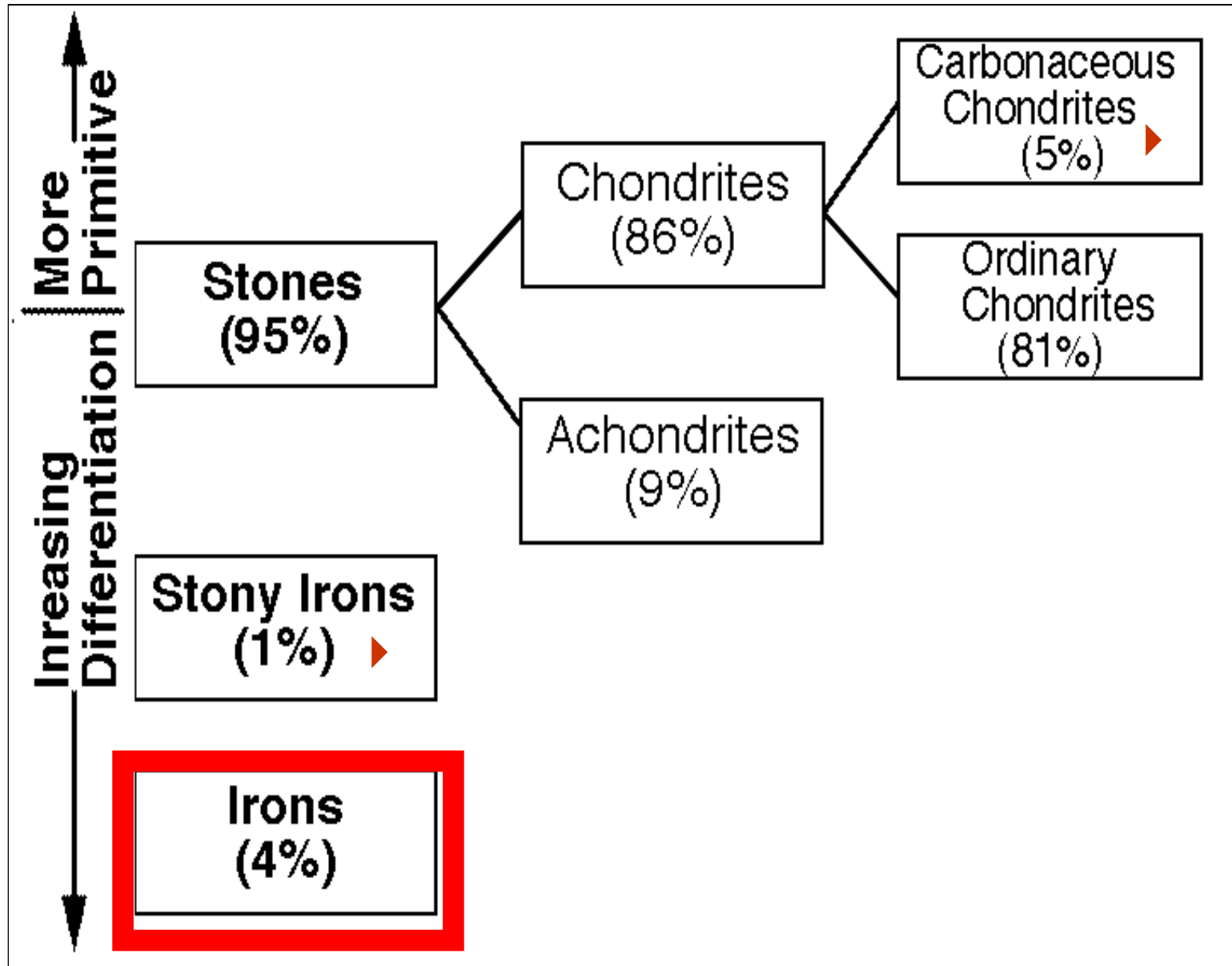
CONNECTIONS ASTÉROÏDES-MÉTÉORITES



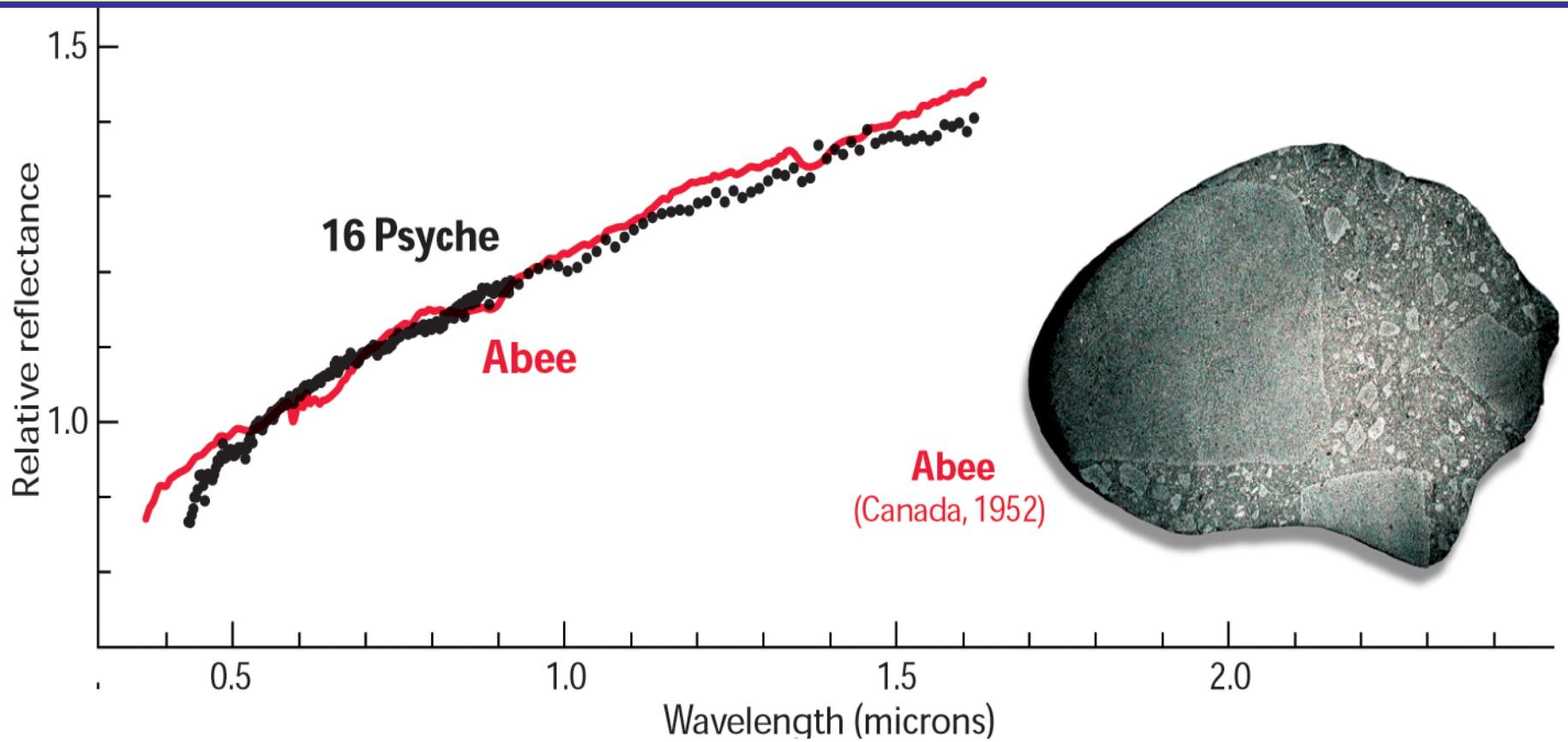
Possible agreement between astéroïdes of type A and stony-iron meteorites.



CONNECTIONS ASTÉROÏDES-MÉTÉORITES

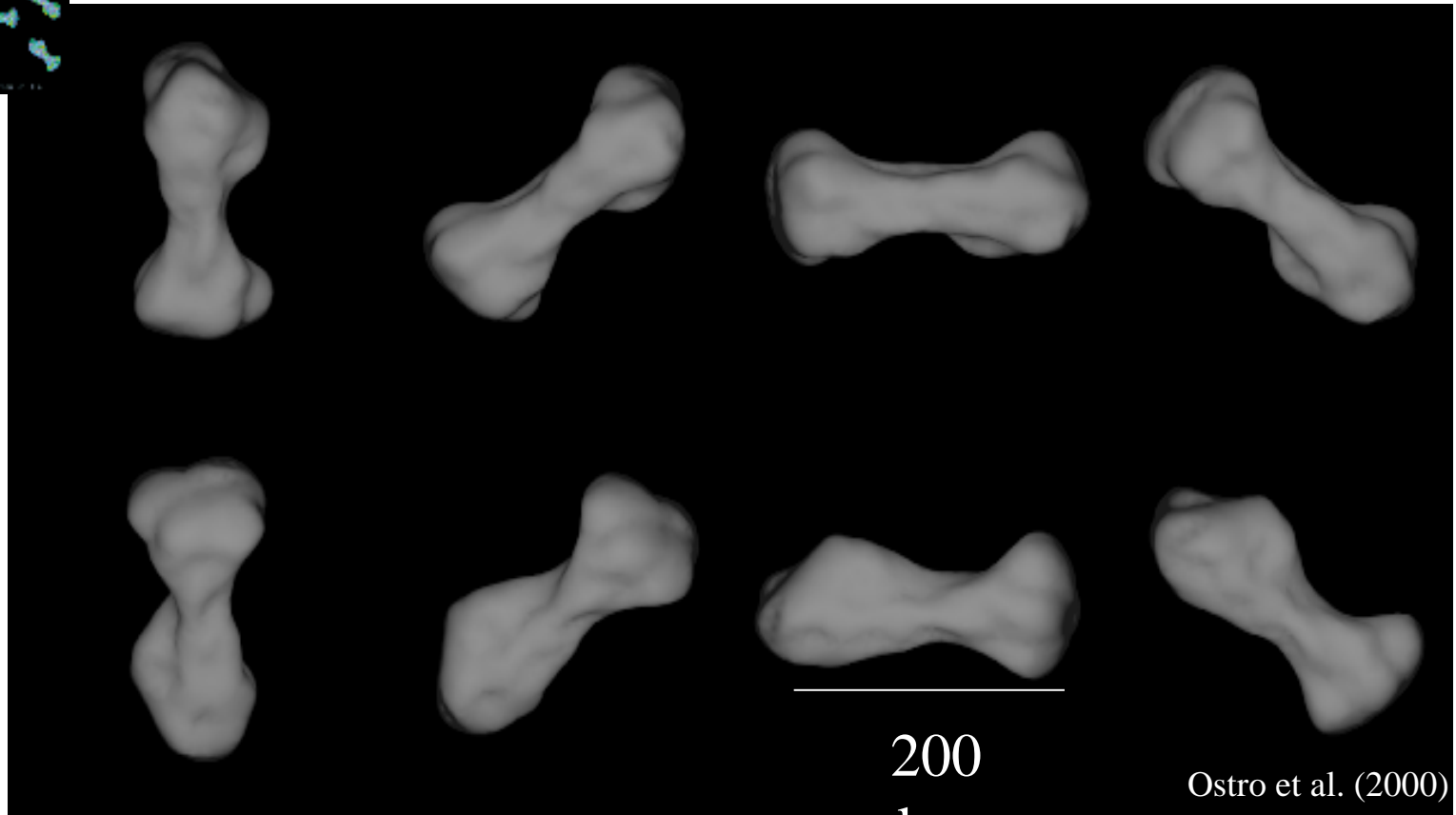


CONNECTIONS ASTÉROÏDES-MÉTÉORITES



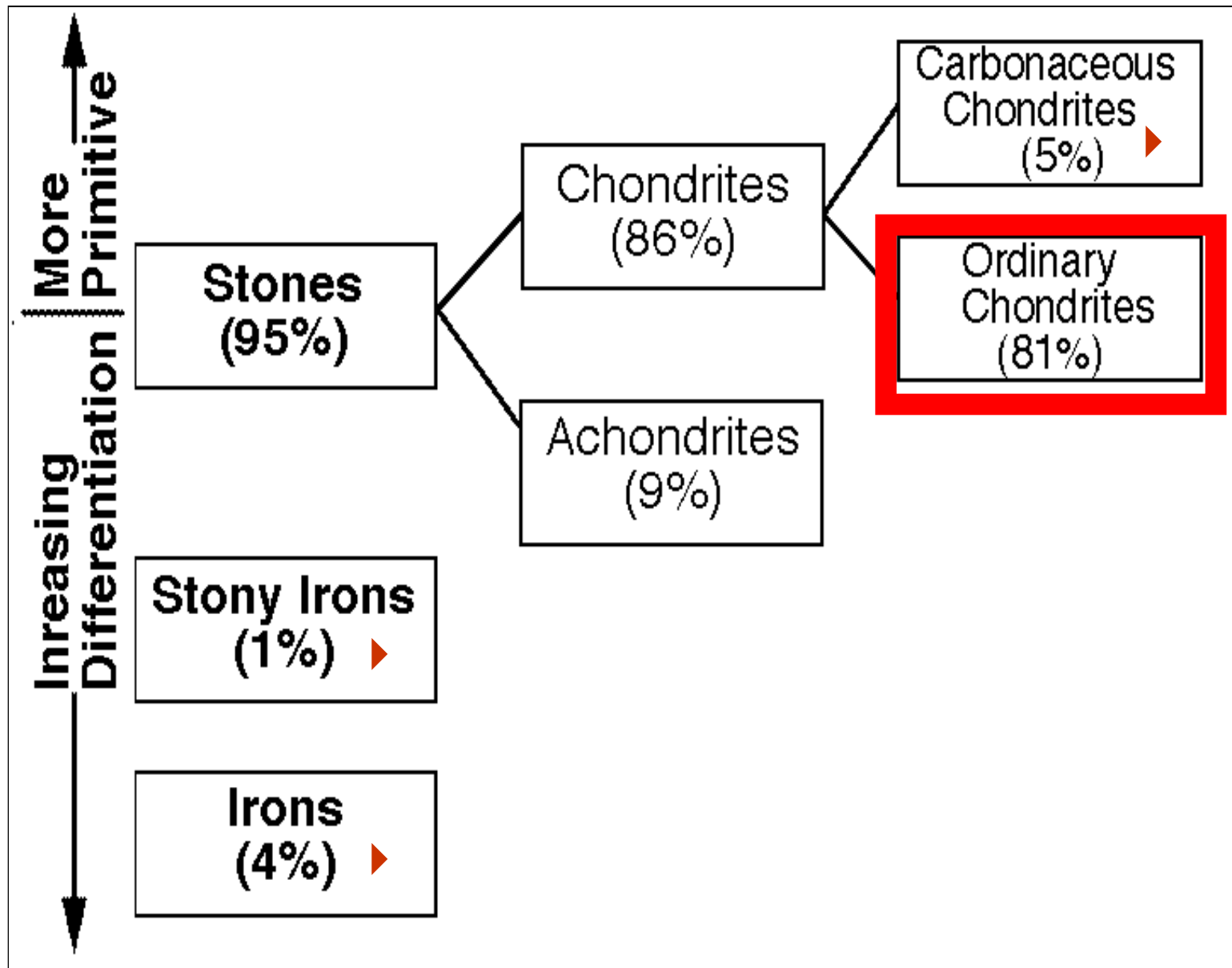
Iron has a smooth spectrum, inclined towards the red.
Quite similar to asteroids of type M.

CONNECTIONS ASTÉROÏDES-MÉTÉORITES

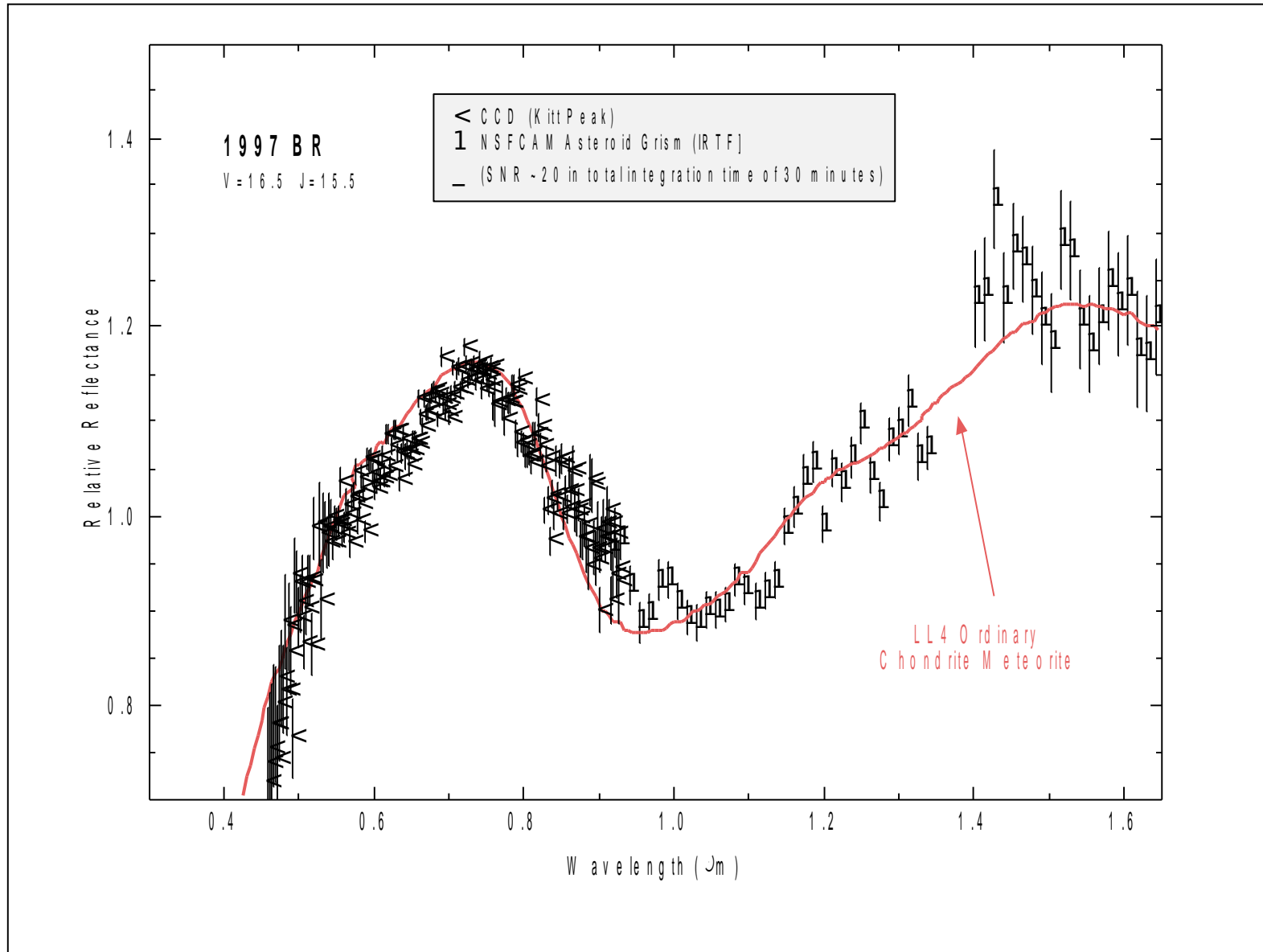


Radar measurements of M-type asteroids provide a strong link to iron meteorite compositions.

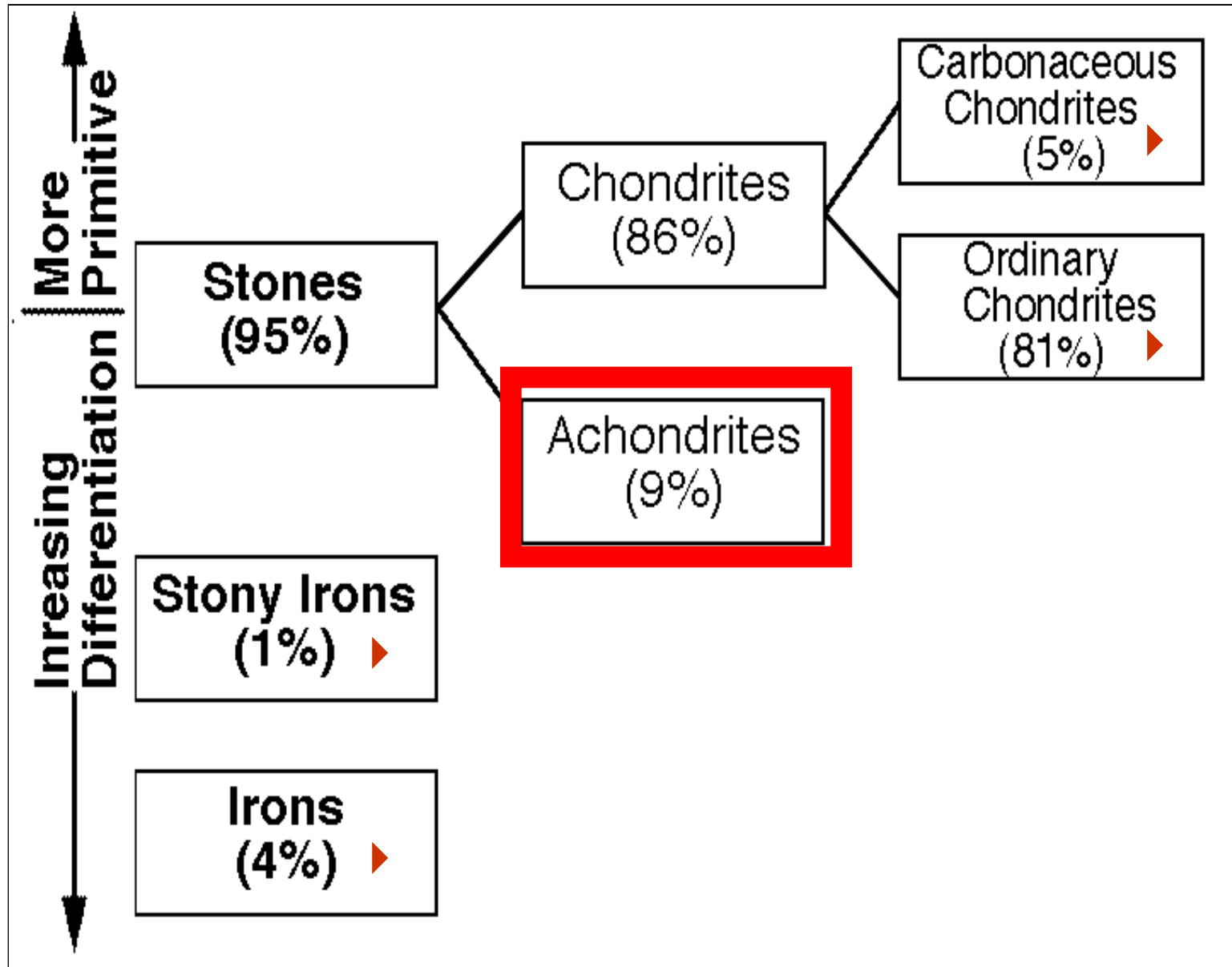
CONNECTIONS ASTÉROÏDES-MÉTÉORITES



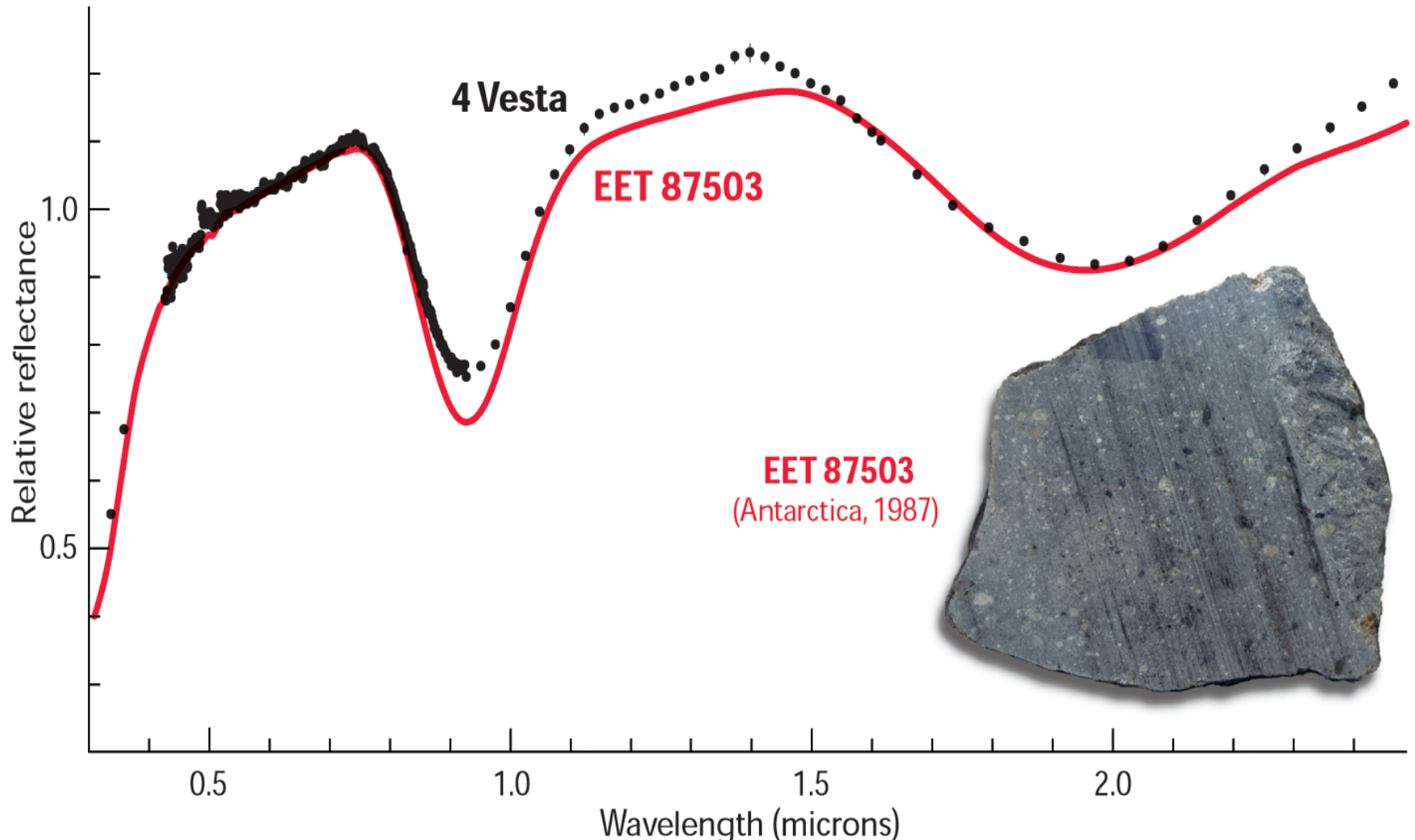
CONNECTIONS ASTÉROÏDES-MÉTÉORITES



CONNECTIONS ASTÉROÏDES-MÉTÉORITES

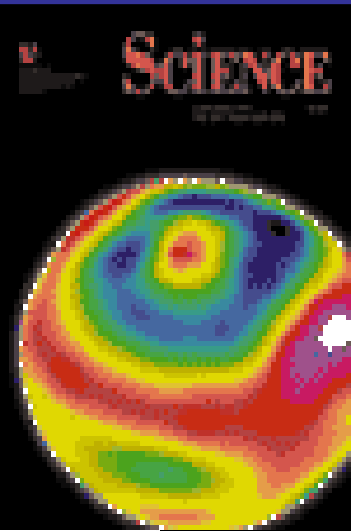


CONNECTIONS ASTÉROÏDES-MÉTÉORITES

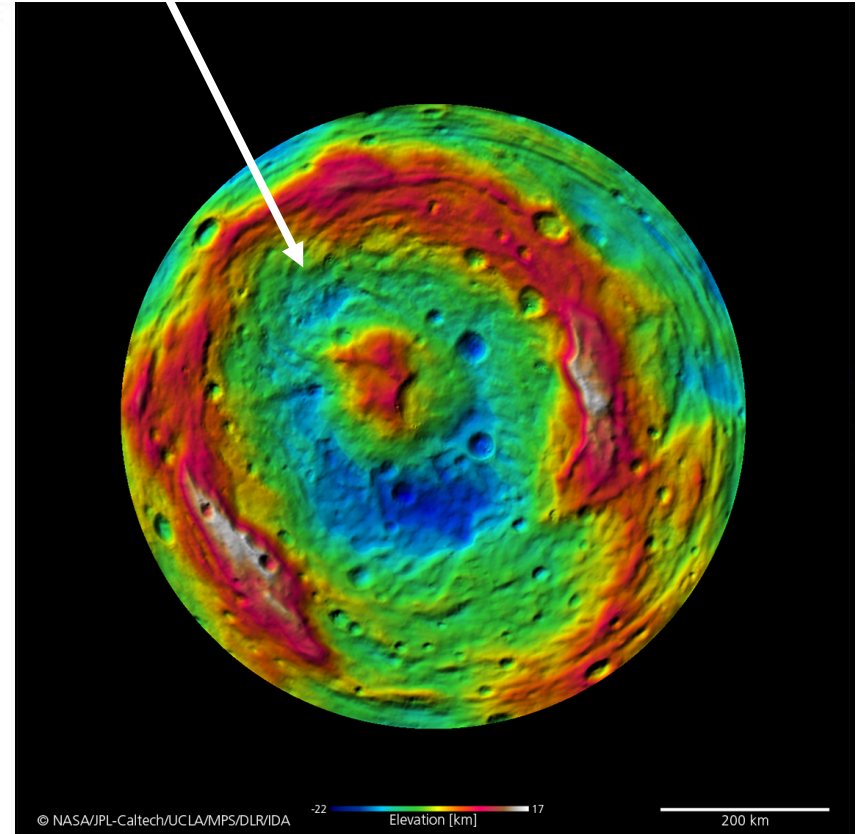
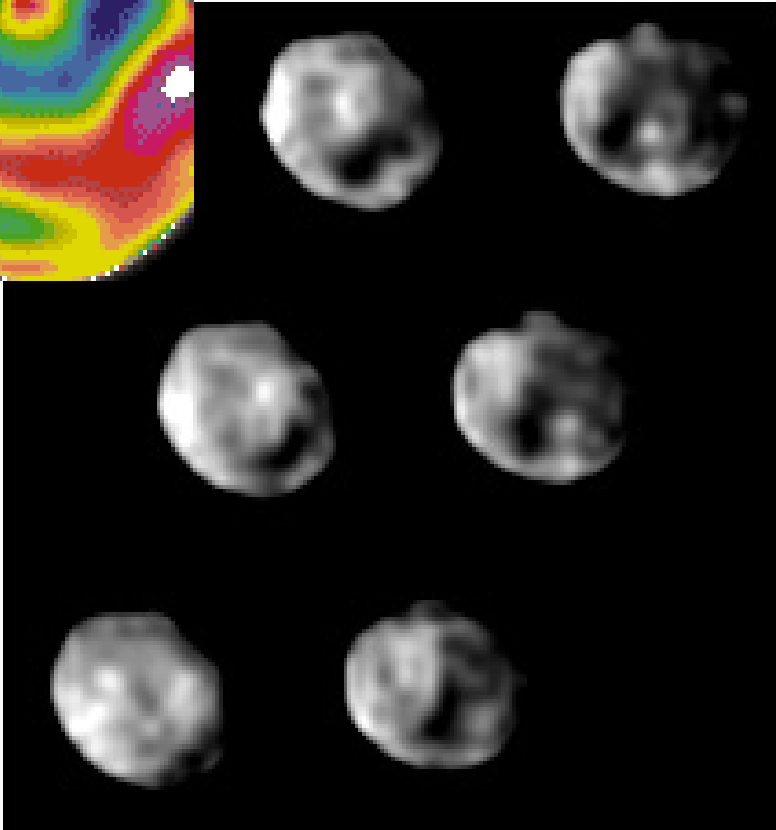


Spectral match between Vesta and achondrite meteorites (HEDs) first found by McCord et al. (1970).

VESTA



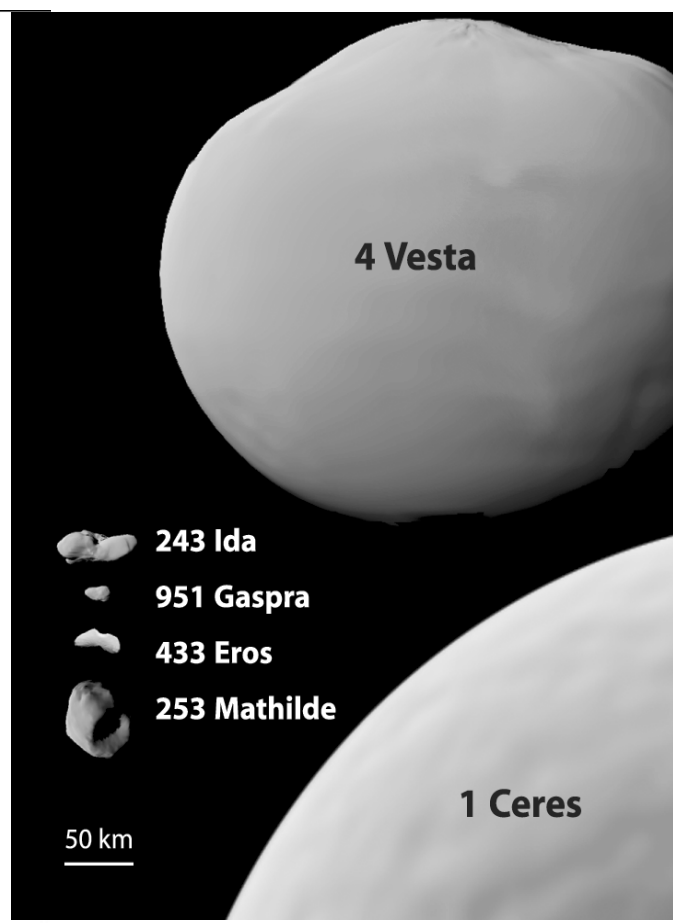
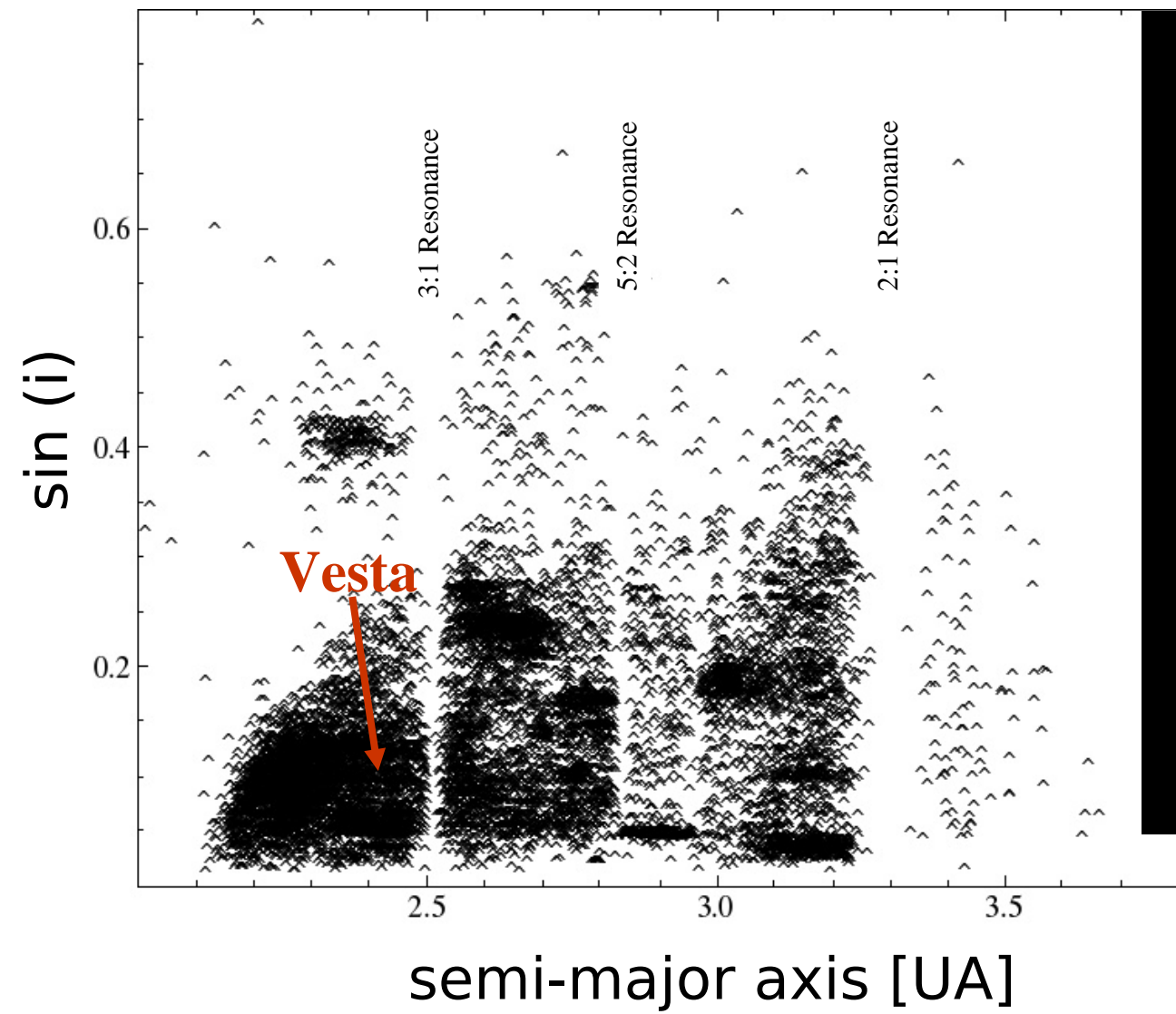
Impact basin of
460 km diameter



Images HST

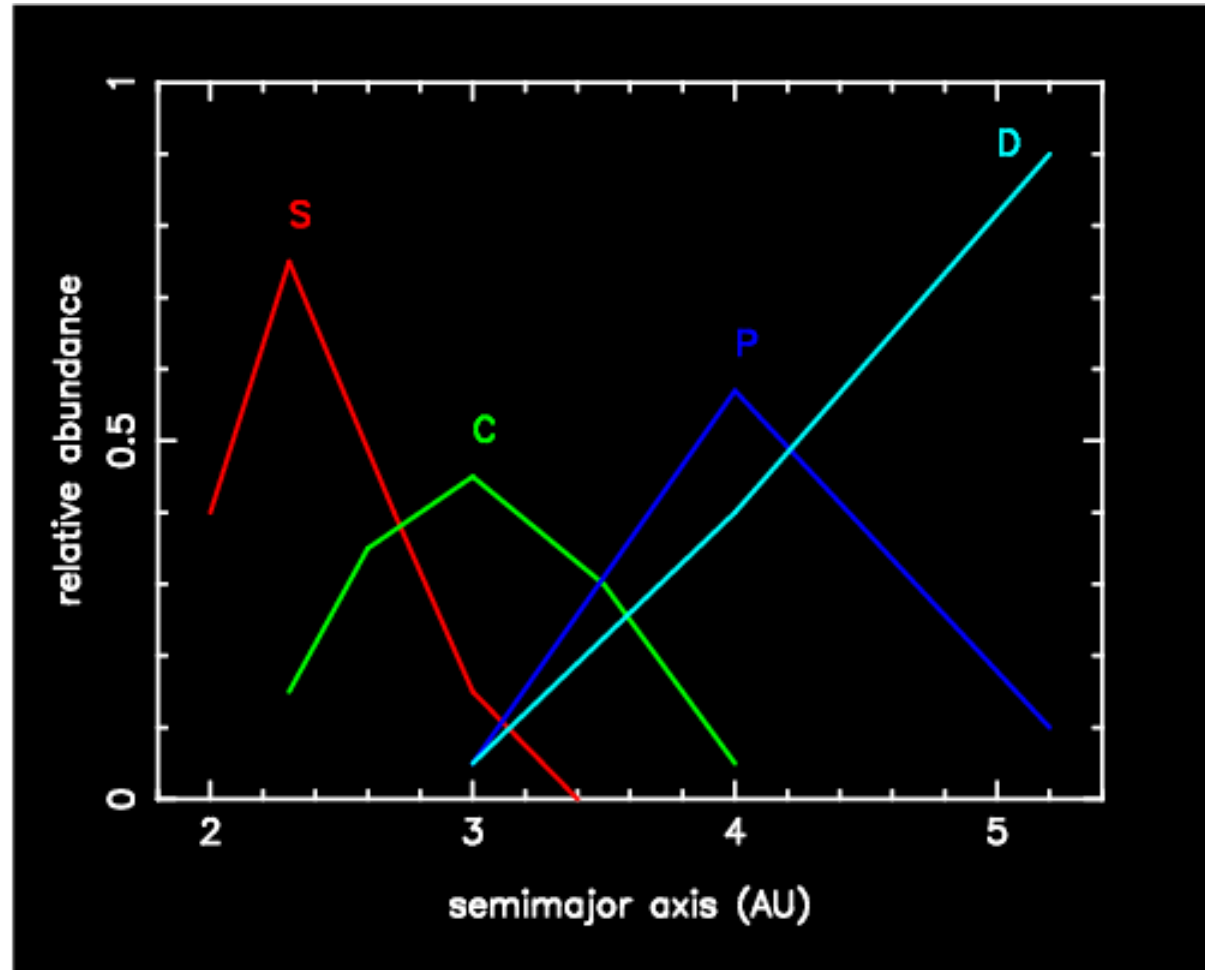
Reconstruction topographique

ASTÉROIDES



ASTÉROIDES

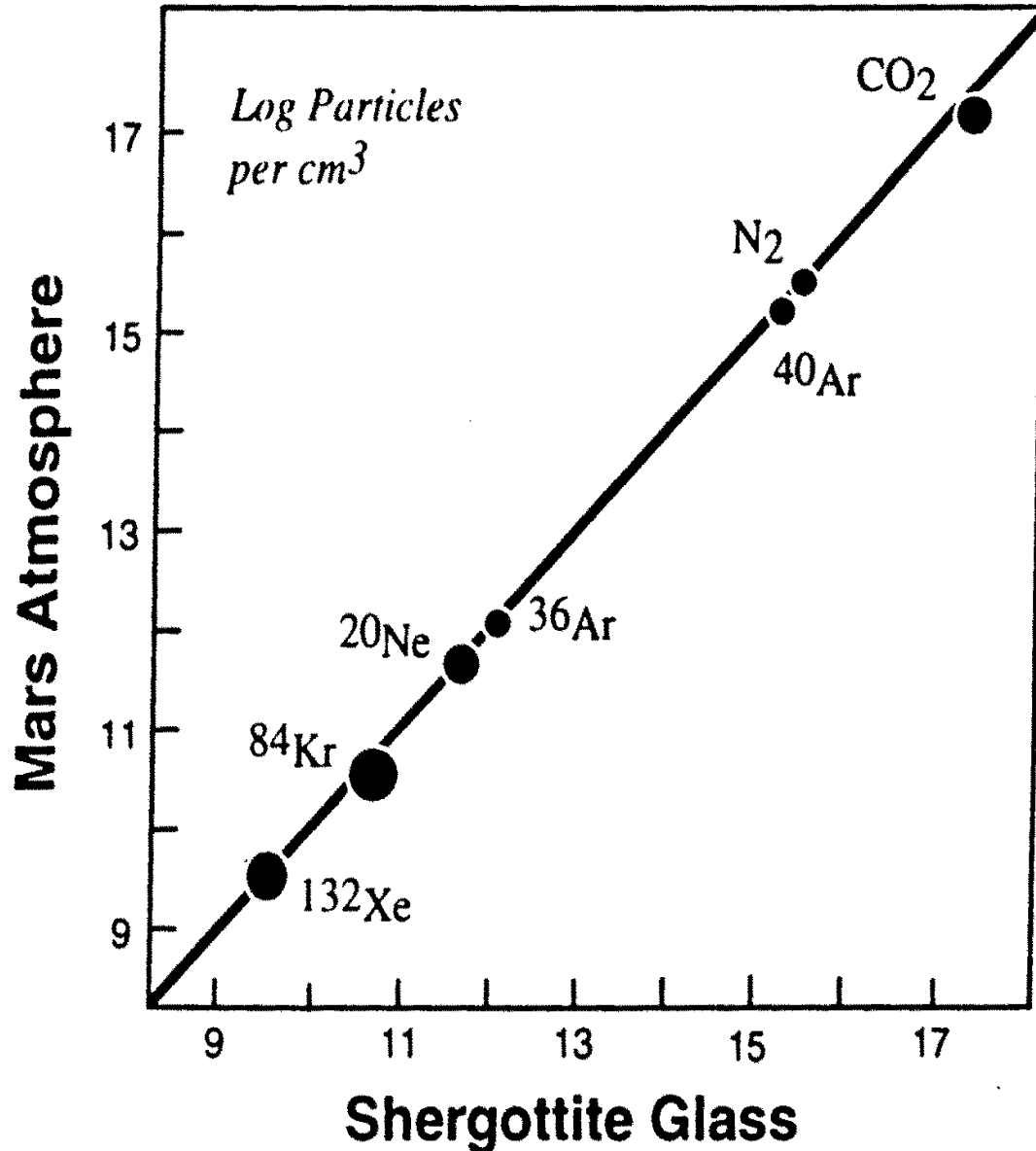
Repartition of taxonomic types :



From Gradie and Tedesco (1982)

MARTIAN METEORITES

Shergottite, Nakhlite, Chassignite (SNC) Meteorites:



Abundances

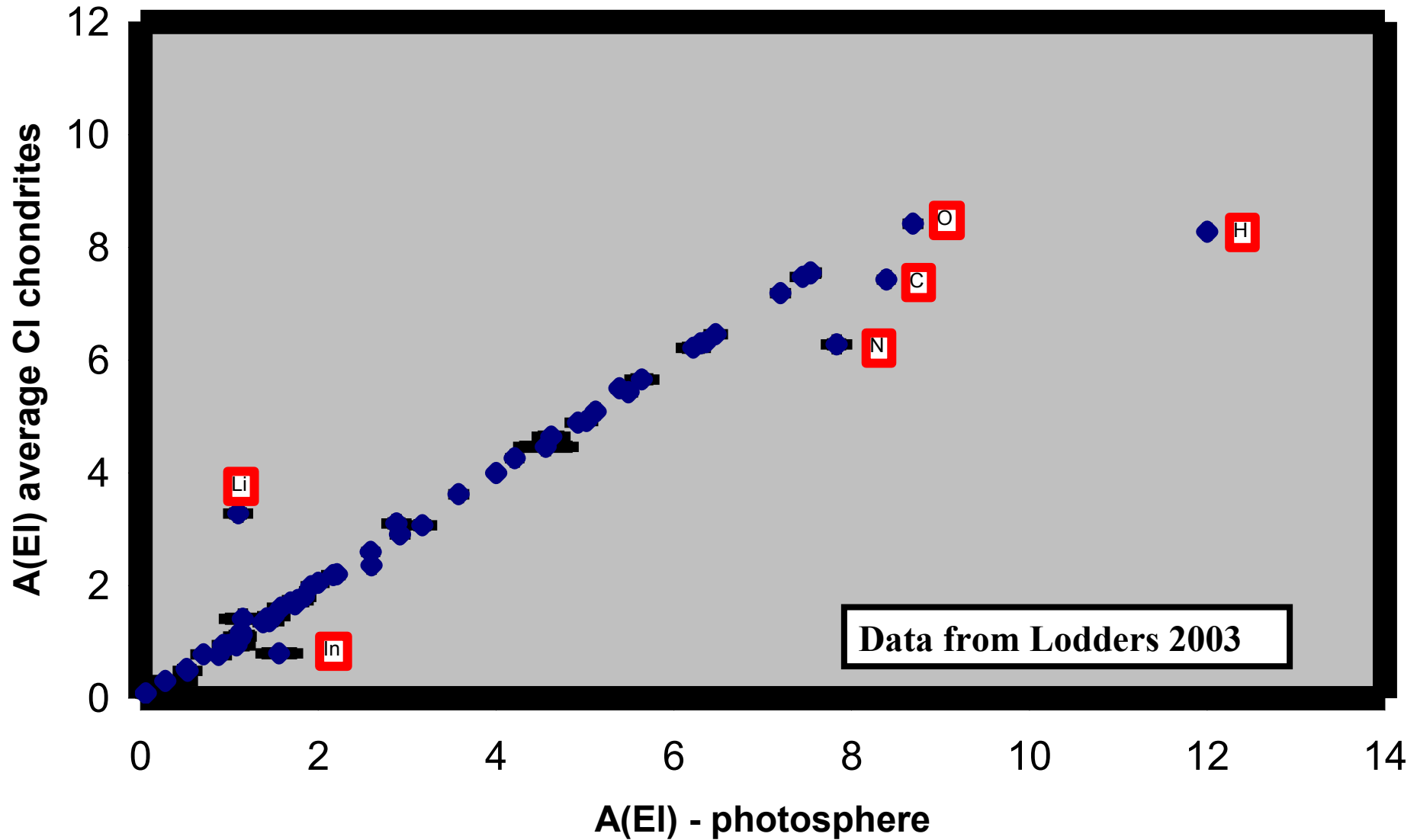
ABUNDANCES

- ★ The abundance of an element EI in the photosphere of the Sun is measured via emission lines.
- ★ In the solar spectroscopic community:

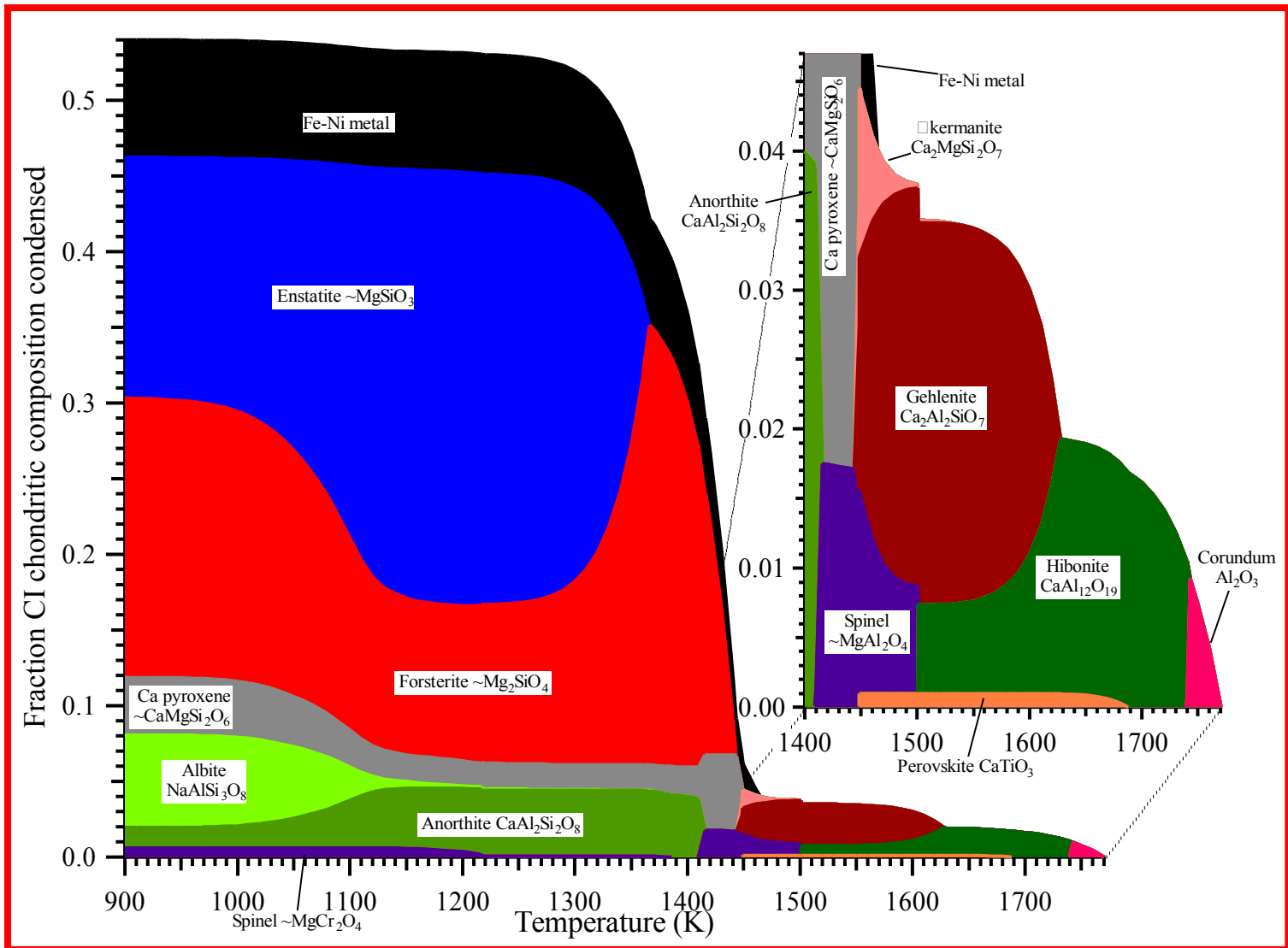
$$A(EI) = \log[n(EI)/n(H)] + 12$$

- ★ In this scale, $A(H) = 12$.
- ★ Meteorites abundances are measured in the laboratory
[more precise]
- ★ Meteorites abundances are rescaled so that
$$A(\text{Si})_{\text{meteorites}} = A(\text{Si})_{\text{photosphere}}$$

ABNUDANCES



CONDENSATION SEQUENCE



Age of Chondrites

RADIOACTIVE DECLINE

☆ Radioactivity discovered by Becquerel in 1896

☆ Radioactivity α : ${}^A_Z\text{P} \rightarrow {}^{A-4}_{Z-2}\text{F} + {}^4\text{He}$

☆ Radioactivity β^- : ${}^A_Z\text{P} \rightarrow {}^A_{Z+1}\text{F} + e^- + \underline{\nu}_e$ (n \rightarrow p + e^- + $\underline{\nu}_e$)

☆ Radioactivity β^+ : ${}^A_Z\text{P} \rightarrow {}^A_{Z-1}\text{F} + e^+ + \bar{\nu}_e$ (n \rightarrow p + e^+ + $\bar{\nu}_e$)

$$\mathbf{N(t) = N(0) \times \exp(-\lambda t)}$$

$$\mathbf{or: N(t) = N(0) \times (1/2)^{-t/T}}$$

☆ Radioactive decay characterized by:

☆ The decay constant (λ)

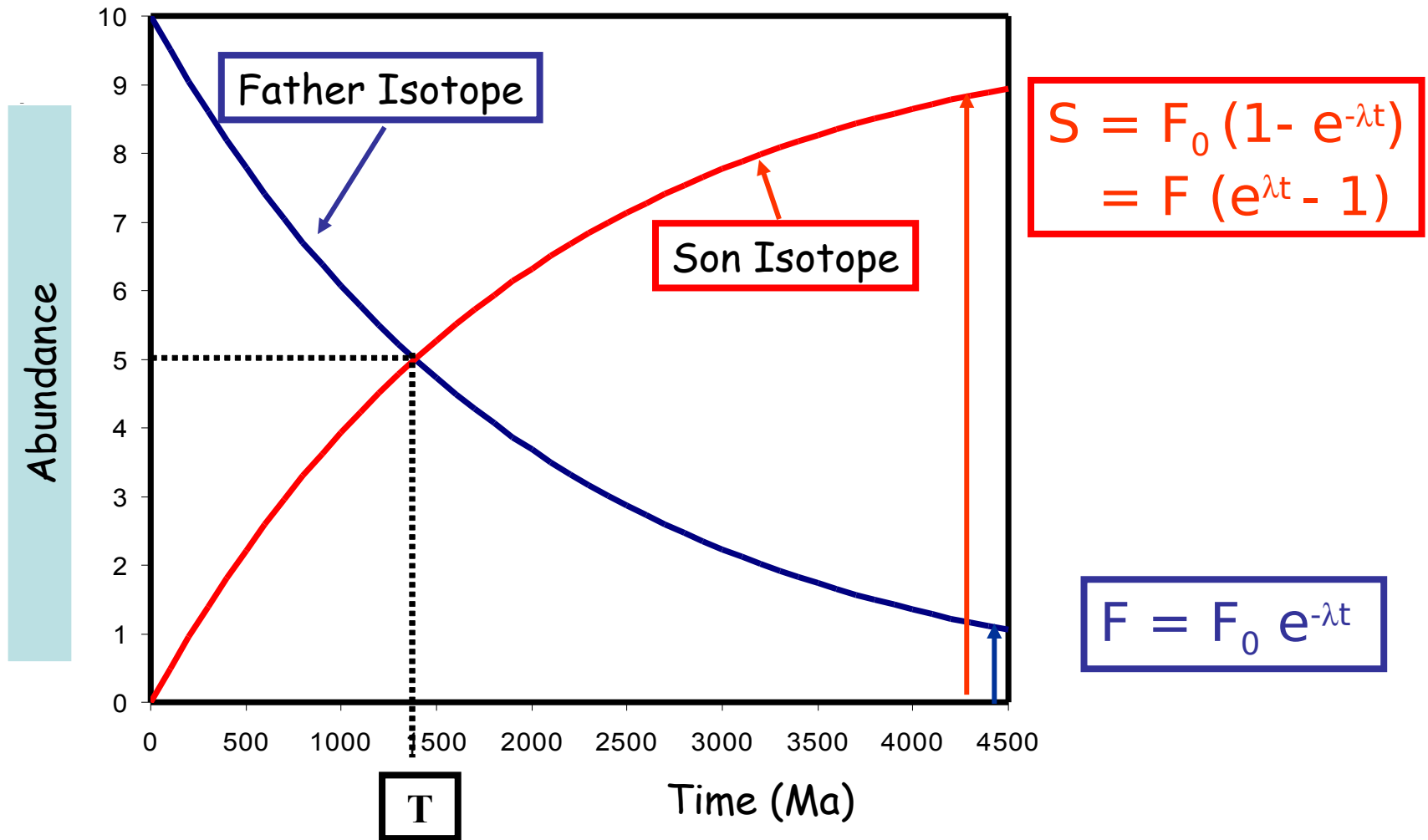
☆ The half-life ($T_{1/2}$)

AGES and ISOTOPIC COSMOCHEMISTRY

One of the most important successes of isotopic cosmochemistry is the ability to date rocks, events...

- ★ Radioactive chronometers are used to date events and processes
- ★ chronometers
 - ★ Long-lived chronometers are still alive ($T_{1/2} > 200 \text{ Ma}$)
 - ★ Short-lived chronometers are extinct (Al-Mg; Hf-W;...)
- ★ Main long-lived chronometers are
 - ★ ^{238}U decaying into ^{206}Pb ($T_{1/2} = 4.5 \times 10^9 \text{ yr}$) **fission**
 - ★ ^{235}U decaying into ^{207}Pb ($T_{1/2} = 7.03 \times 10^8 \text{ yr}$) **fission**
 - ★ ^{87}Rb decaying into ^{87}Sr ($T_{1/2} = 4.8 \times 10^{10} \text{ yr}$) **β^-**

PRINCIPLES of DATATION



If F_0 is known, and $S=0$ at $t=0$, one just needs to measure F or S to deduce t . But in general, neither F_0 nor $S(t=0)$ are known...

ABSOLUTE DATATION: long period isotopes

The system Pb-Pb :

$$^{206}\text{Pb} = ^{206}\text{Pb}_0 + ^{238}\text{U}(e^{\lambda_{238}t} - 1)$$

avec $\lambda_{238} = 1,55125 \cdot 10^{-10} \text{ an}^{-1}$ (*demi-vie = 4,47 Ga*)

$$^{207}\text{Pb} = ^{207}\text{Pb}_0 + ^{235}\text{U}(e^{\lambda_{235}t} - 1)$$

avec $\lambda_{235} = 9,8485 \cdot 10^{-10} \text{ an}^{-1}$ (*demi-vie = 0,704 Ga*)

$$\frac{^{206}\text{Pb}}{^{204}\text{Pb}} = \frac{^{206}\text{Pb}_0}{^{204}\text{Pb}} + \frac{^{238}\text{U}}{^{204}\text{Pb}}(e^{\lambda_{238}t} - 1)$$

$$\frac{^{207}\text{Pb}}{^{204}\text{Pb}} = \frac{^{207}\text{Pb}_0}{^{204}\text{Pb}} + \frac{^{235}\text{U}}{^{204}\text{Pb}}(e^{\lambda_{235}t} - 1)$$

ABSOLUTE DATATION: long period isotopes

The system Pb-Pb :

(Hyp : $^{238}\text{U}/^{235}\text{U} = C^{\text{te}} = 137,88$)

$$\frac{\left(\frac{^{207}\text{Pb}}{^{204}\text{Pb}}\right) - \left(\frac{^{207}\text{Pb}}{^{204}\text{Pb}}\right)_0}{\left(\frac{^{206}\text{Pb}}{^{204}\text{Pb}}\right) - \left(\frac{^{206}\text{Pb}}{^{204}\text{Pb}}\right)_0} = \frac{1}{137,88} \frac{e^{\lambda_{235}t} - 1}{e^{\lambda_{238}t} - 1}$$

or: $(y-y_0)/(x-x_0) = a(t)$

Equation defining a straight line of slope depending on time :
the slope gives the age ! This straight line is an **isochrone**.

For an isochrone, at least 2 points are needed, from 2
minerals of a same object, having different initial U/Pb.

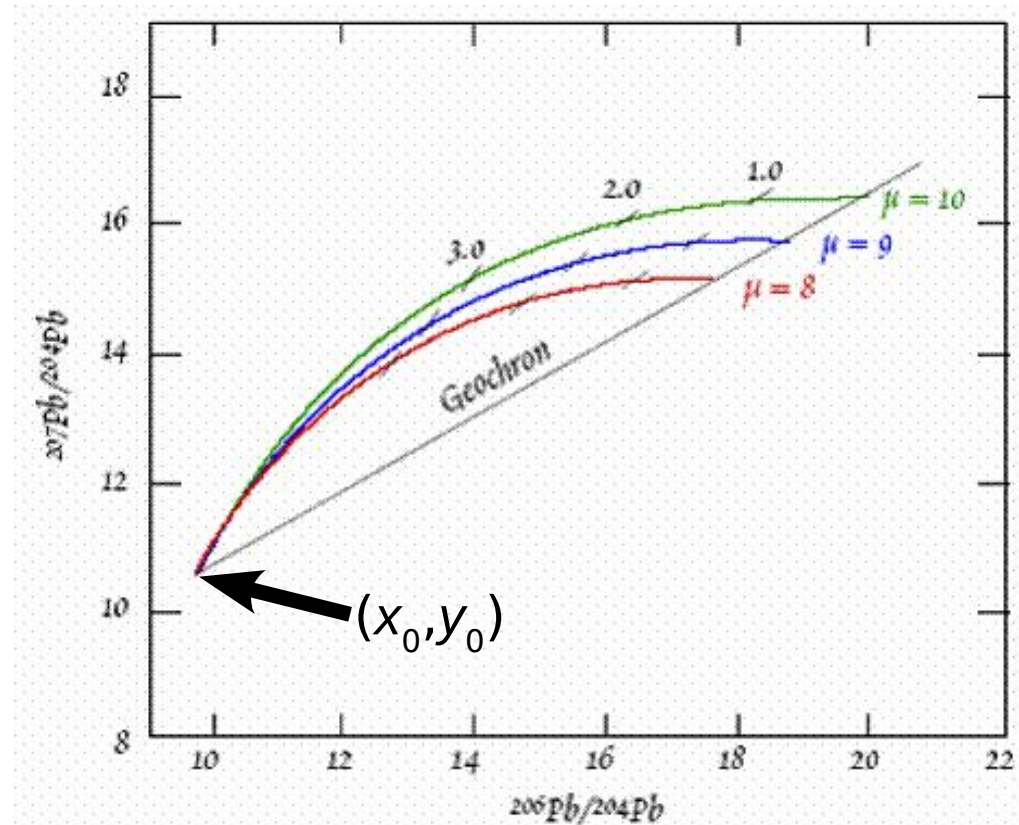
ABSOLUTE DATATION: long period isotopes

The system Pb-Pb :

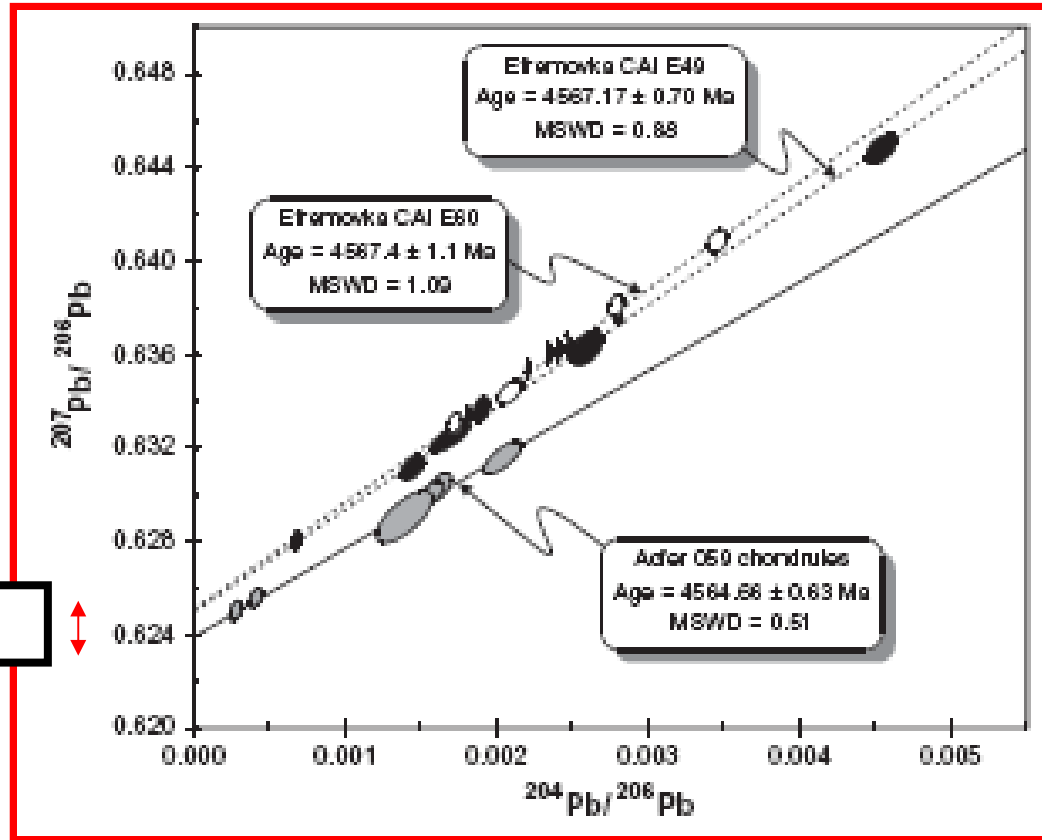
Consider the formation of an asteroid. N kinds of minerals are created, with various $\mu = U/Pb$ (chemistry).

Then, the system doesn't evolve chemically anymore, but $U \rightarrow Pb$ inside the minerals.

In the Pb-Pb diagram, the N points are aligned on a line of slope a (age of the asteroid).



ABSOLUTE DATATION: CAI & CHONDRULES



- ☆ The oldest dated CAI has 4567.17 ± 0.70 Ma
 - ☆ Compatible with 4566 ± 2 Ma (Manhès et al. 1988)
- ☆ Does that date the origin of Solar System?

RELATIVE DATATION: short period isotopes

If extinct radioactivity, $F=0$, $S=S_0+F_0$.

Ex: The system Hf-W ($t_{1/2} \sim 9$ Myr) :

Hafnium 182 decomposes into Tungsten 182.

$${}^{182}\text{W} = {}^{182}\text{W}_{init} + {}^{182}\text{Hf}_{init}$$

$$\underbrace{\left(\frac{{}^{182}\text{W}}{{}^{184}\text{W}} \right)}_y = \underbrace{\left(\frac{{}^{182}\text{W}}{{}^{184}\text{W}} \right)}_b_{init} + \underbrace{\left(\frac{{}^{182}\text{Hf}}{{}^{180}\text{Hf}} \right)}_a_{init} * \underbrace{\left(\frac{{}^{180}\text{Hf}}{{}^{184}\text{W}} \right)}_x$$

The higher the rate ${}^{182}\text{Hf}/{}^{180}\text{Hf}$ was, the more the proportion of ${}^{182}\text{W}$ inside the tungsten increases with the general Hf/W ratio.

RELATIVE DATATION: short period isotopes

The system Al-Mg ($t_{1/2} \sim 0.71$ Ma) : extinct radioactivity.

$$\left(\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}} \right) = \left(\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}} \right)_{t=t_f} + \left(\frac{{}^{26}\text{Al}}{{}^{24}\text{Mg}} \right)_{t=t_f} \quad (t_f = \text{formation time})$$

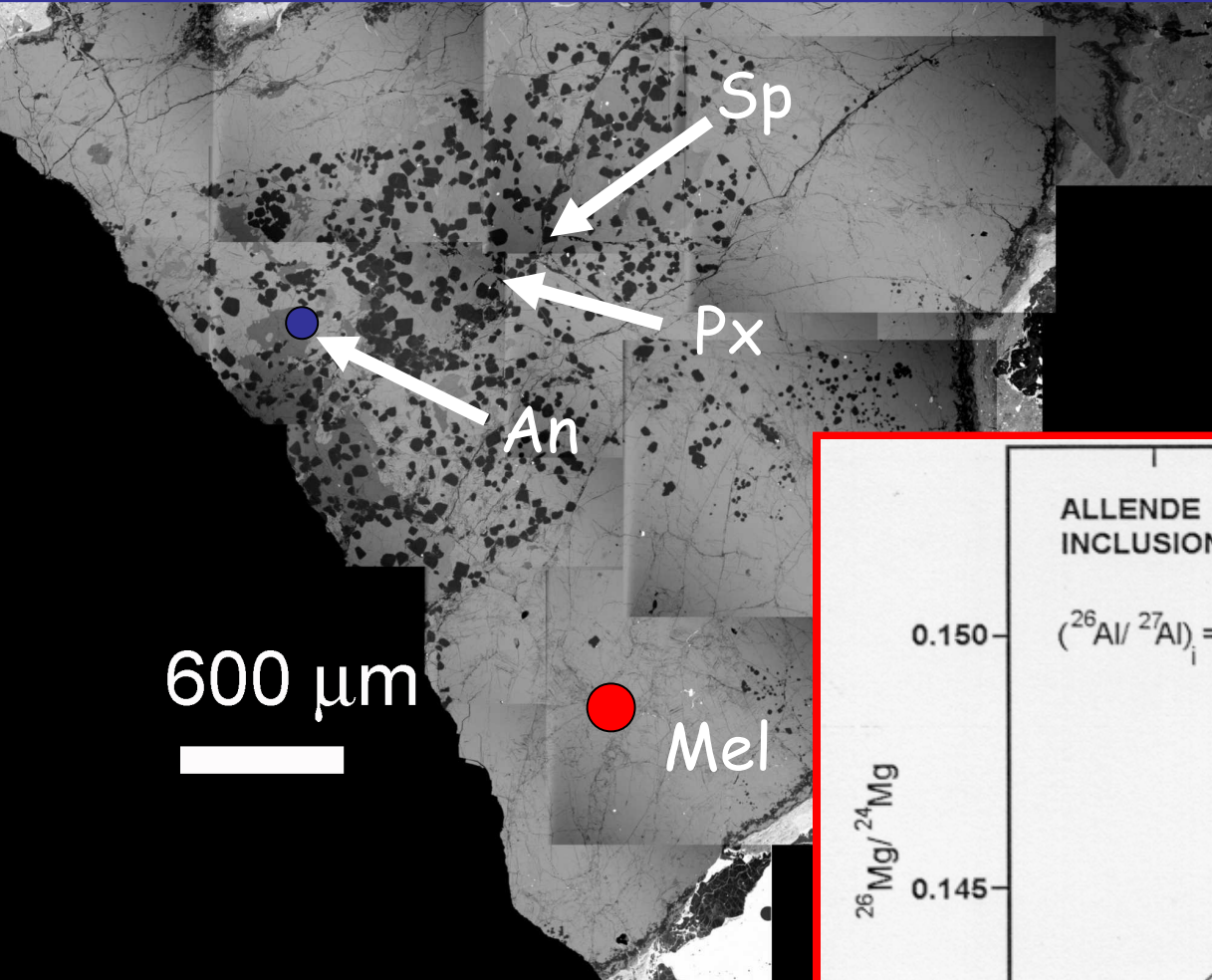
$$\left(\frac{{}^{26}\text{Al}}{{}^{24}\text{Mg}} \right)_{t=t_f} = \left(\frac{{}^{26}\text{Al}}{{}^{27}\text{Al}} \right)_{t=t_f} \left(\frac{{}^{27}\text{Al}}{{}^{24}\text{Mg}} \right) \quad \left(\frac{{}^{26}\text{Al}}{{}^{27}\text{Al}} \right)_{t=t_f} = \left(\frac{{}^{26}\text{Al}}{{}^{27}\text{Al}} \right)_{t=t_0} e^{-\lambda(t_f-t_0)}$$

$$\underbrace{\left(\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}} \right)}_y = \underbrace{\left(\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}} \right)_{t=t_f}}_b + \underbrace{\left(\frac{{}^{26}\text{Al}}{{}^{27}\text{Al}} \right)_{t=t_0}}_a e^{-\lambda(t_f-t_0)} * \underbrace{\left(\frac{{}^{27}\text{Al}}{{}^{24}\text{Mg}} \right)}_x$$

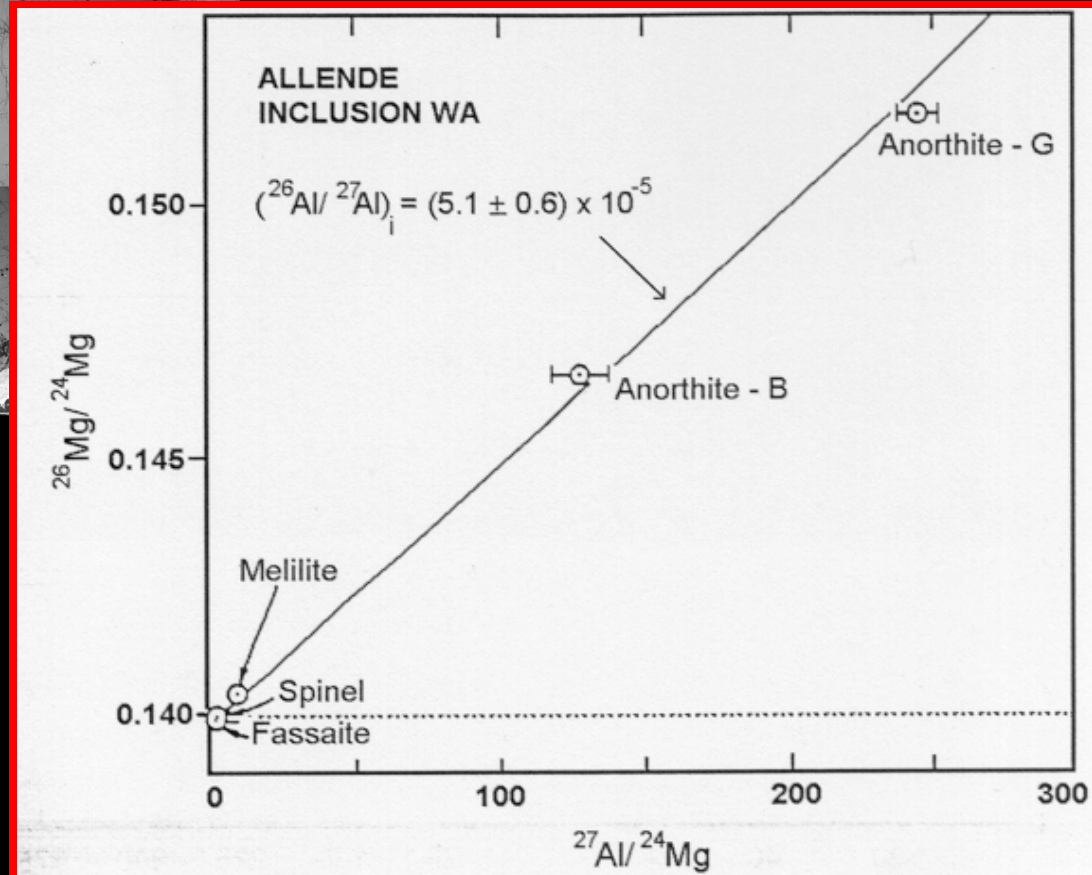
If one measures x and y of several minerals of a same object, the points align on a straight line of slope a : an **isochrone**.

Same slope $a \iff$ same age t_f (under the assumption that $({}^{26}\text{Al}/{}^{27}\text{Al})_{t_0}$ is the same in all the Solar System). **EXO**

RELATIVE DATATION: short period iso<topes



Melilite
Anorthite
Spinel
Pyroxene

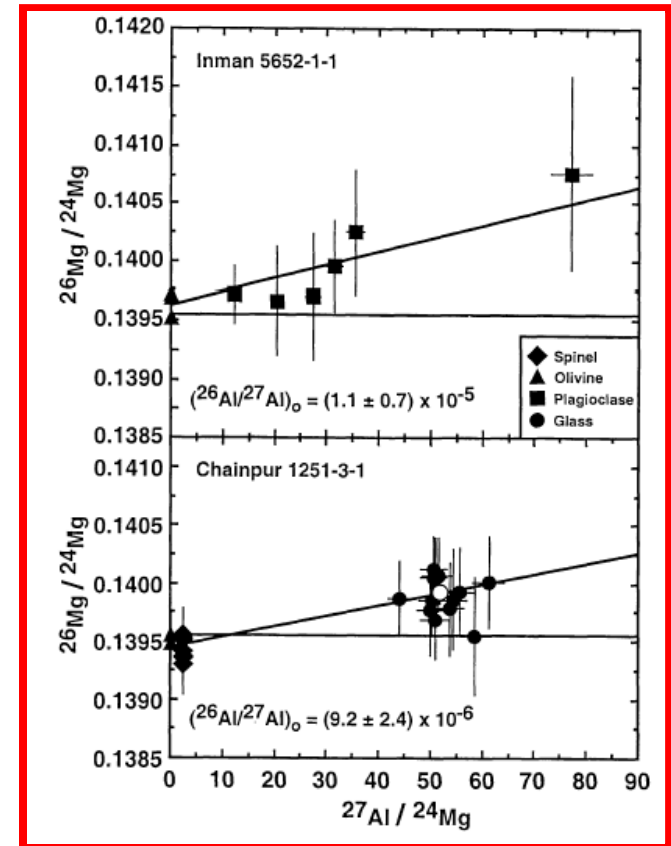
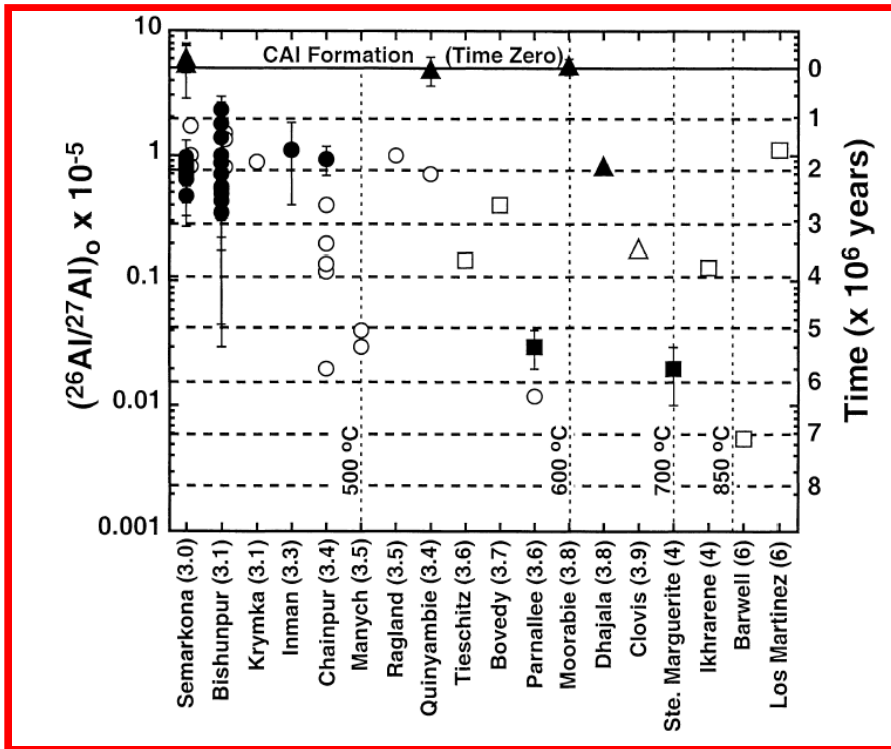


CAI MRS6 (Leoville, CV3)
Crystallized **4.567 Ga**
(BSE image)

^{26}Al inside CHONDRULES

Maximum for $a = (^{26}\text{Al}/^{27}\text{Al})_{t_0} e^{-\lambda(t-t_0)}$ inside chondrules : $\sim 10^{-5}$.

Inside CAI : $\sim 4,5 \times 10^{-5}$.



^{26}Al inside CHONDRULES

Maximum for $a = (^{26}\text{Al}/^{27}\text{Al})_{t_0} e^{-\lambda(t-t_0)}$ inside chondrules : $\sim 10^{-5}$.

Inside CAI : $\sim 4,5 \times 10^{-5}$.

Thus $\exp(-\lambda(t_{\text{CAI}} - t_0)) / \exp(-\lambda(t_{\text{chondres}} - t_0)) = 4,5$

so $\exp(-\lambda t_{\text{CAI}}) = 4,5 * \exp(-\lambda t_{\text{chondres}})$

so $t_{\text{CAI}} = t_{\text{chondres}} - \ln(4,5)/\lambda$

Conclusion: chondrules formed 2 Myrs after the CAIs.

We set $t_{\text{CAI}} = t_0 = 0$, birth of the Solar System.

PRINCIPLES of DATATION

Fundamental assumptions :

- The isotopic distribution of the chemical elements was the same at $t=0$ everywhere in the Solar System.
- Chemical fractionation doesn't induce any isotopic fractionation.

Attention :

- The ages obtained are “closure ages” (i.e. the time for which the system hasn't exchanged elements with the outer world).
- In fact, datation is made from isochrones (so from the analysis of several chemical components of the sample), thus one date the age of ***chemical fractionation*** of the system.

THE Hf – W SYSTEM : DIFFERENTIATION

The system Hf-W ($t_{1/2} \sim 9 \text{ Ma}$) :

Hafnium 182 declines into Tungstène 182.

Similarly as for the Al-Mg, system, one finds :

$$\underbrace{\left(\frac{{}^{182}\text{W}}{{}^{184}\text{W}} \right)}_y = \underbrace{\left(\frac{{}^{182}\text{W}}{{}^{184}\text{W}} \right)}_b_{CAI} + \underbrace{\left(\frac{{}^{182}\text{Hf}}{{}^{180}\text{Hf}} \right)}_a_{CAI} e^{-\lambda(t-t_{CAI})} * \underbrace{\left(\frac{{}^{180}\text{Hf}}{{}^{184}\text{W}} \right)}_x$$

But Hafnium is lithophile (goes into the mantle)
while the Tungsten is siderophile (goes into the core) :
chemical fractionation.

One date the closure of the system “mantle” or “core”, that is
the ***differentiation*** of a planet !

THE Hf – W SYSTEM : DIFFERENTIATION

The system Hf-W ($t_{1/2} \sim 9 \text{ Ma}$) :

Late differentiation: all the ^{182}Hf has declined, all the W in the core.

Fast differentiation : ^{182}Hf in the mantle, produces there some ^{182}W .

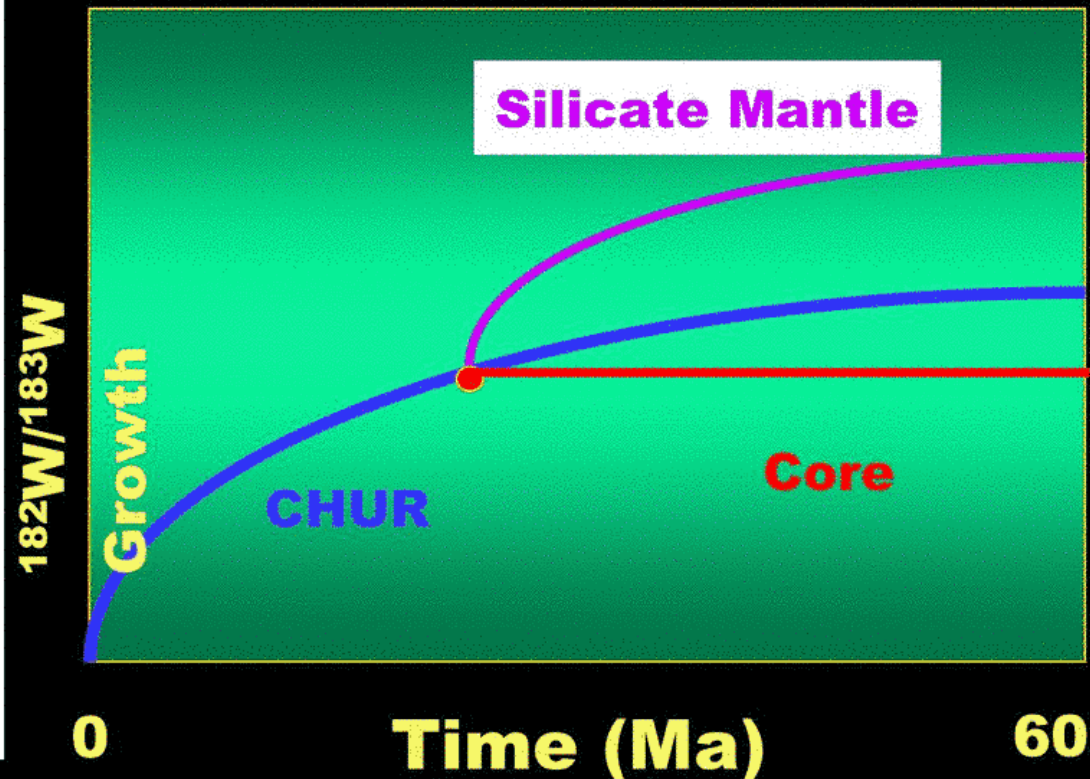
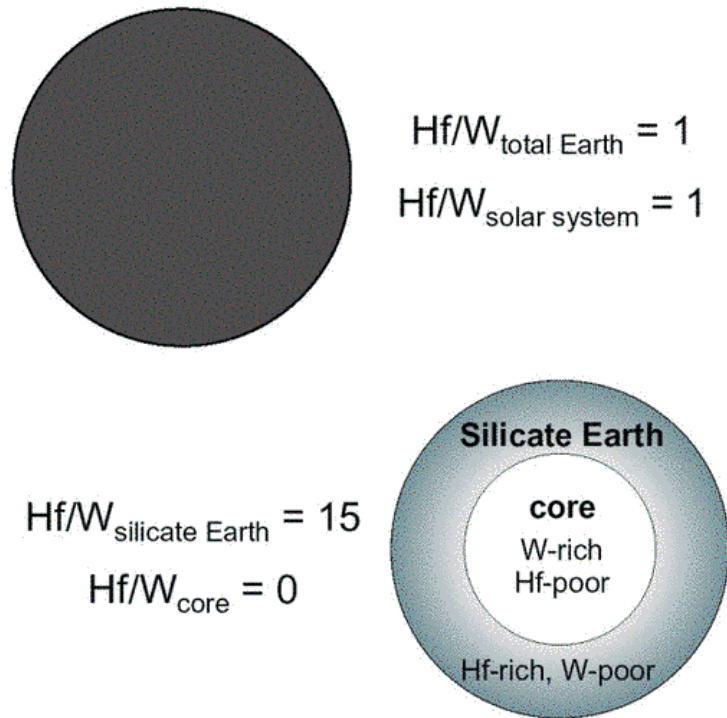


Fig. 2. Schematic showing the fractionation of Hf from W in the early solar system.

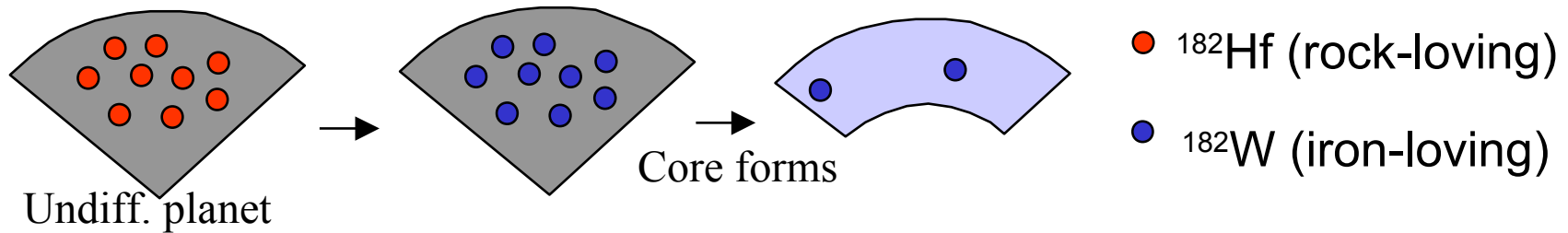
After Halliday et al. (2000)

THE Hf – W SYSTEM : DIFFERENTIATION

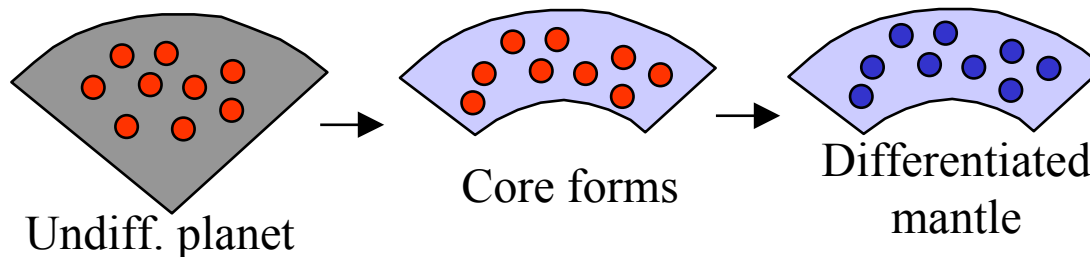
Age of the Moon, through $^{182}\text{Hf} / ^{182}\text{W}$ chronometer :

$^{182}\text{Hf} \rightarrow ^{182}\text{W}$, half life $\tau = 9 \cdot 10^6$ years.

Late core formation : no excess ^{182}W in the mantle.



Early core formation : excess ^{182}W in the mantle.



(sketch courtesy:
F. Nimmo)

Excess ^{182}W in lunar rocks \rightarrow Moon solidified within 60 Myrs
(Kleine et al. 2005, Lee et al. 1997, 2002)

THE Hf – W SYSTEM : DIFFERENTIATION

Definition :

$$\epsilon_{^{182}\text{W}} = \left[\frac{\left(\frac{^{182}\text{W}}{^{184}\text{W}} \right)_{\text{objet}}}{\left(\frac{^{182}\text{W}}{^{184}\text{W}} \right)_{\text{reference}}} - 1 \right] * 10^4$$

Analyse of 3 kinds of lunar rocks (Kleine et al, 2007, Science):

	KREEP	Low W basalt	High W basalt
$\epsilon^{182}\text{W}$:	0,06 +/- 0,20	1,18 +/- 0,20	2,14 +/- 0,57
$^{180}\text{Hf}/^{184}\text{W}$:	10 +/- 10	26,5 +/- 6,5	> 45

EXERCICE: Deduce the age of solidification of the Moon.
One gives $(^{182}\text{W}/^{184}\text{W})_{\text{ref}} = 1$, $(^{182}\text{Hf}/^{180}\text{Hf})_{\text{CAI}} = 10^{-4}$.

THE Hf – W SYSTEM : DIFFERENTIATION

Unluckily, ^{182}W is also coming from ^{181}Ta capturing a neutron (cosmic rays).

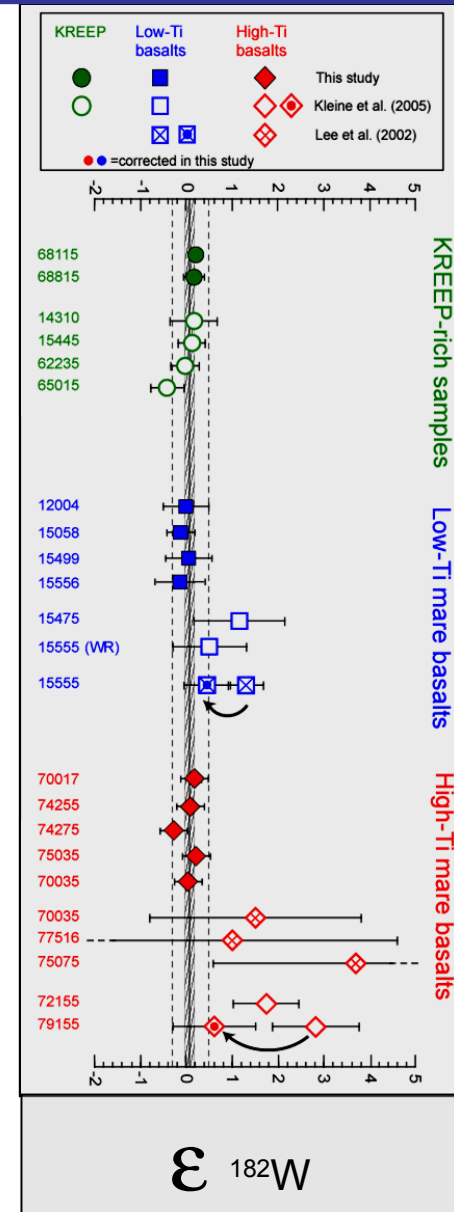
Touboul et al. (2007, Nature) : “The dominant ^{182}W component in most lunar rocks reflects cosmogenic production”

New data from unpolluted samples :

“lunar and terrestrial mantles have identical $^{182}\text{W}/^{184}\text{W}$. This [...] constrains the age of the Moon and Earth to 62 Myrs (+90/-10)”.

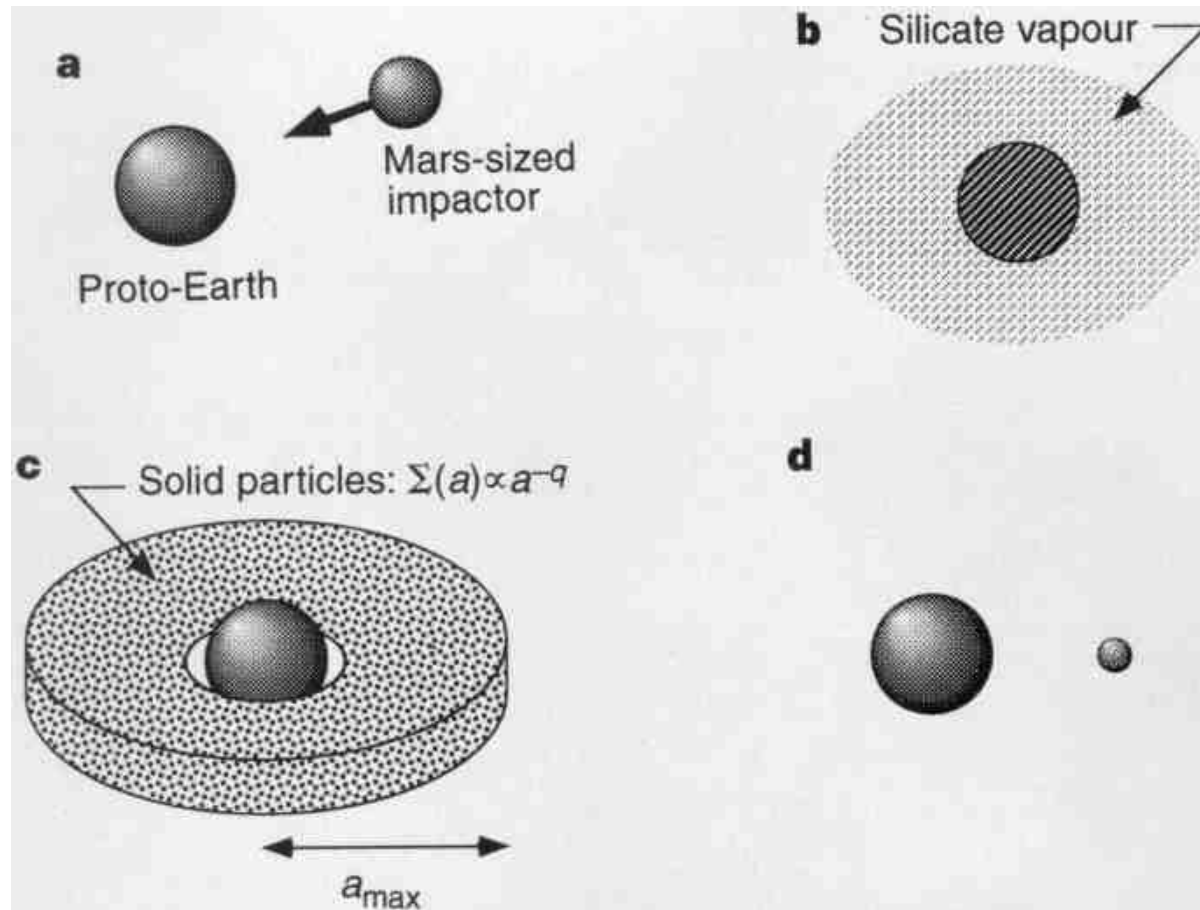
Touboul et al. (2008) :

- Same result for plagioclase separates from two ferroan anorthosites (i.e. crust).
- Measure of basalts cosmogenic pollution.



MOON FORMATION

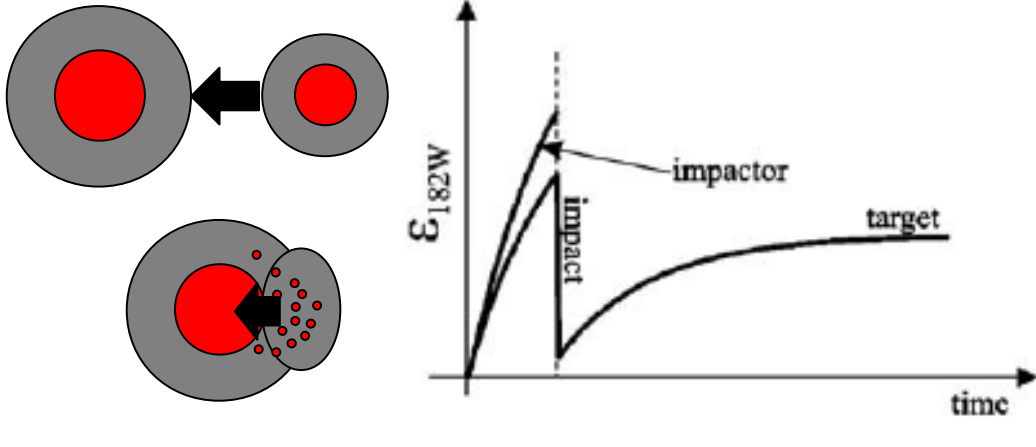
The Moon formed by a giant impact on the proto-Earth of a proto-planet of the size of Mars.



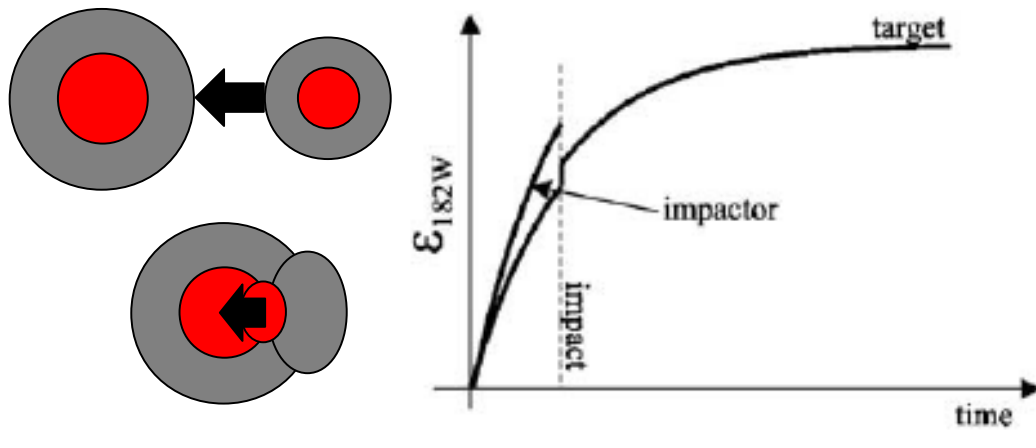
Do we date the differentiation of the proto-Earth ?
That of the impactor ? The collision ?

MOON FORMATION

Re-equilibration of the mantles :



Core merging :



(sketches courtesy: F. Nimmo)

It depends...

If reequilibration, the anomaly in ^{182}W is reduced (or reset) by the impact.

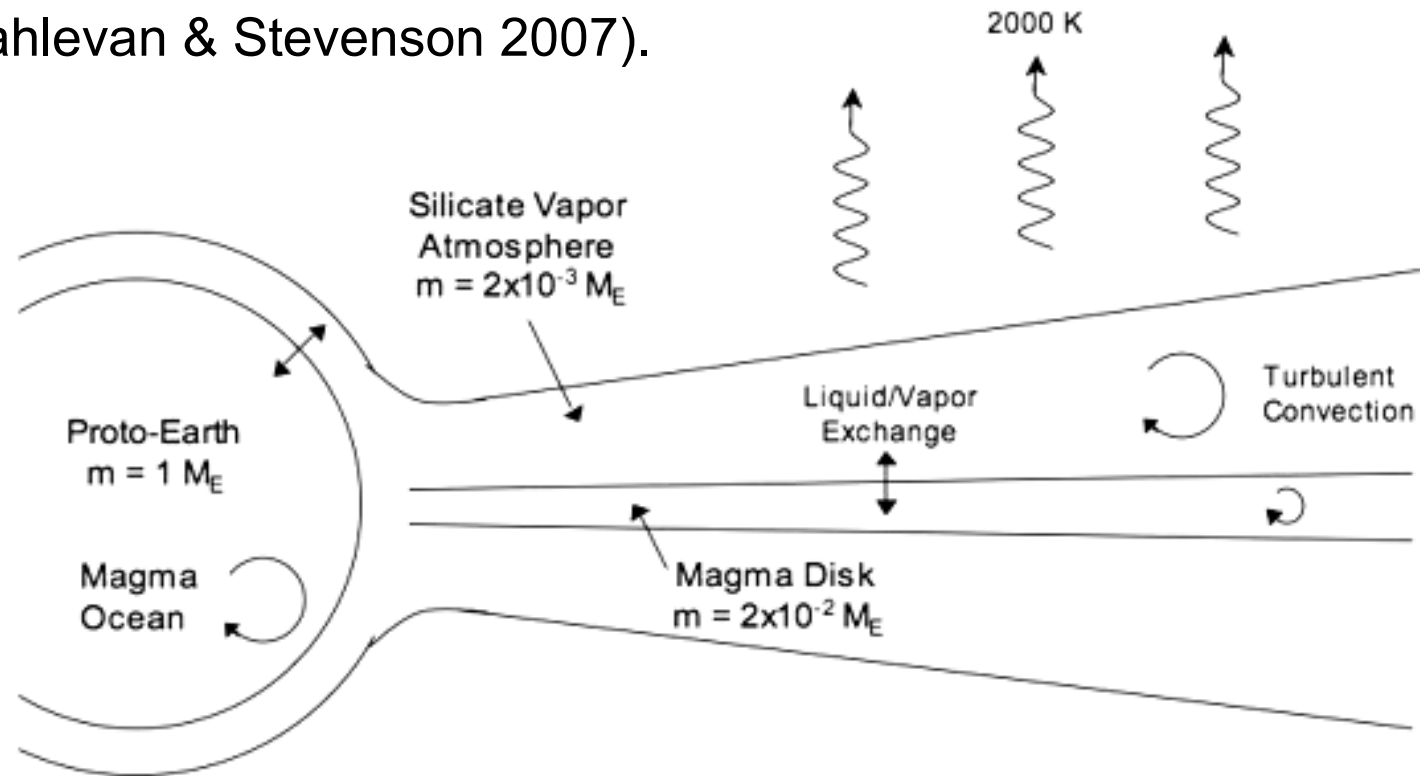
Total reequilibration -> one dates the impact.

No reequilibration : we get the average $\epsilon^{182}\text{W}$.

Observations of terrestrial planets and asteroids suggest that there has been many giant impacts after 10 Ma, with reequilibration (Nimmo & Agnor 2006).

MOON FORMATION

Isotopic equilibration just after the impact, between a magma ocean Earth, and a molten disc, via an atmosphere of silicates (Pahlevan & Stevenson 2007).



Disc during 100 to 1000 years, at 2500 K, 10 to 100 Bar.

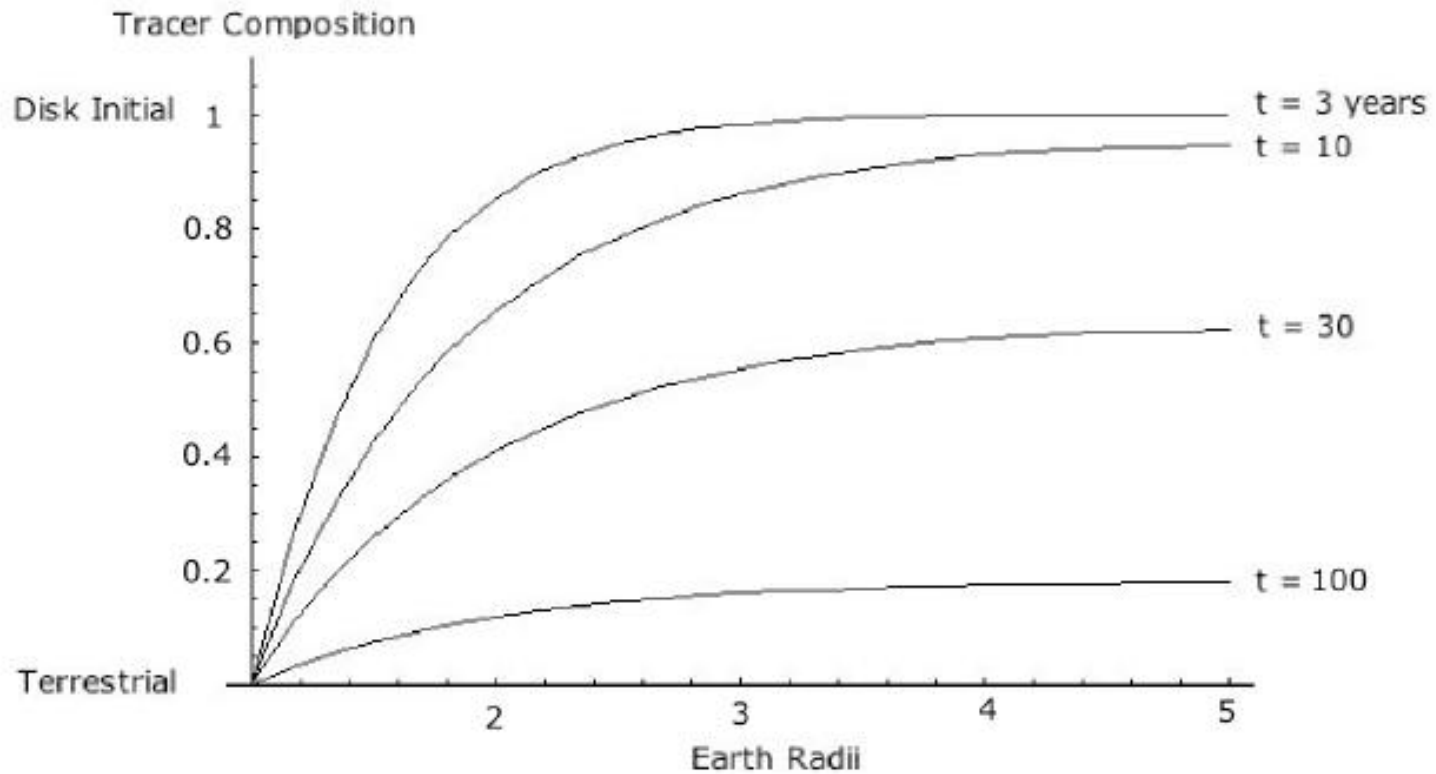
Typical exchange time between vapour and liquid : week.

Convection in the ocean : week. In the disc or the gas : day.

MOON FORMATION

Problem : radial mixing.

It needs more than 30 years to diffuse material to 4 terrestrial radii.

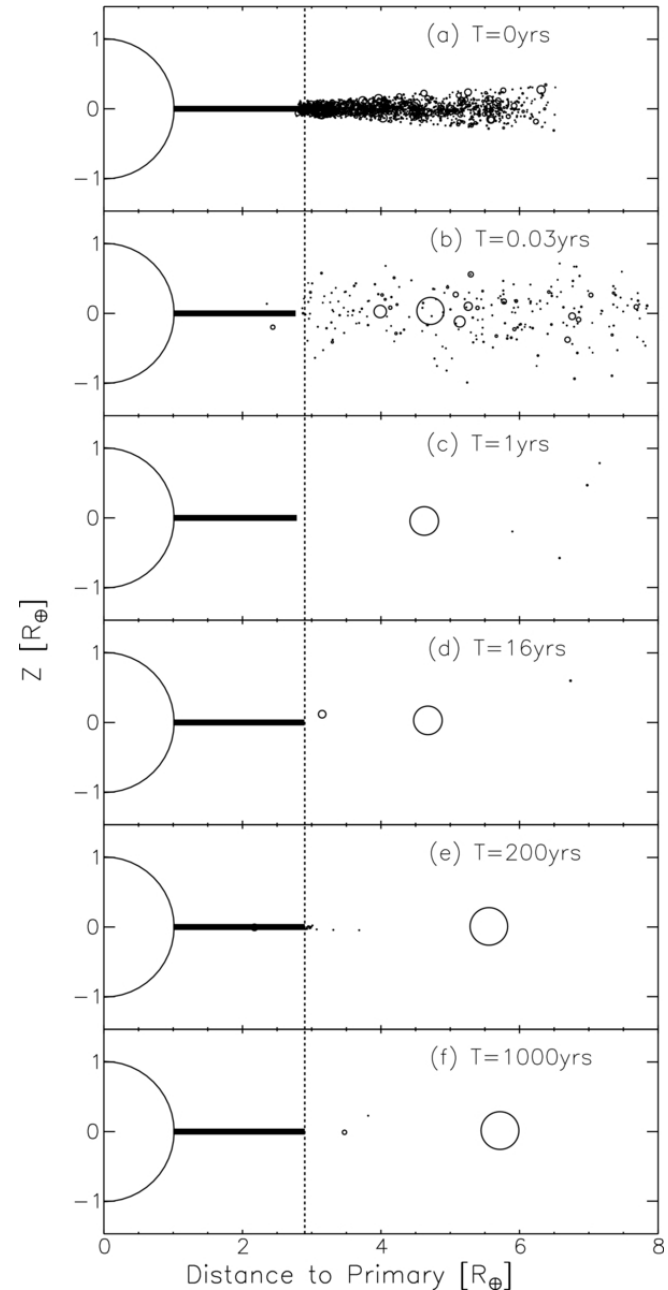


MOON FORMATION

Problem : radial mixing.

It needs more than 30 years to diffuse material to 4 terrestrial radii.

Salmon & Canup (2012) :
Moon accretion stalls for
~100 years because the
growing Moon repels the
disc, and prevents its
spreading.



MOON FORMATION

Conclusion :

The Moon formed after a giant impact on Earth.

This impact caused an Earth-Moon isotopic equilibration.

As the mantles of the Earth and the Moon still have the same $^{182}\text{W}/^{184}\text{W}$, ^{182}Hf was extinct at that time : the impact took place (at least) 60 Myrs after the CAI.

New result:

Jacobson, Morbidelli, et al, DPS meeting, oct. 2013, Denver

The amount of « late veneer » that the Moon accreted after it formed is correlated with the time of the last giant impact. For reasonable estimates of the late veneer, the impact must be late ($\sim 95 \pm 30$ Myrs).